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Description

This invention relates to novel pyrazoloazoleazomethine dyes, and more particularly to novel pyrazoloazoleazomethine dyes having an improved hue and molecular extinction coefficient.

Known pyrazoloazoleazomethine dyes include various dyes obtained by the coupling reaction of 1H-pyrazolo[1,5-a]benzimidazoles disclosed, e.g., in U.S. Patent 3,061,432, 1H-pyrazolo[1,5-c]-1,2,4-triazoles disclosed, e.g., in U.S.Patent 3,725,067, 1H-imidazo[1,2-b]pyrazoles disclosed, e.g., in U.S. Patent 4,500,630,1H-pyrazolo[1,5-b]-1,2,4-triazoles disclosed, e.g., in U.S. Patent 4,540,654, 1H-pyrazolo[1,5-d]-tetrazoles disclosed, e.g., in Japanese Patent Application (OPI) No. 33552/85 (the term "OPI" as used herein means an "unexamined published patent application"), or 1H-pyrazolo[1,5-b]pyrazoles disclosed, e.g., in Japanese Patent Application (OPI) No. 43659/85 with the oxidation product of an aromatic primary amine color developing agent for photography in the existence of an alkali.

It is disclosed in the above described patents and Japanese Patent Application (OPI) No. 186567/85 that pyrazolotriazoleazomethine dyes can be utilized as, in particular, image-forming magenta dyes for silver halide color photographic materials by utilizing the above-described coupling reaction.

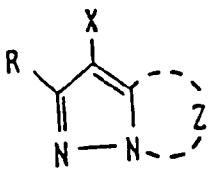
While pyrazoloneazomethine dyes which are utilized as magenta dyes for conventional silver halide color photographic materials have, in addition to main absorptions, harmful side absorptions at the shorter wavelength side of the main absorptions, the above-described pyrazoloazoleazomethine dyes have less side absorptions and hence give a clearer hue as compared to the pyrazoloneazomethine dyes.

Also, some pyrazoloazoleazomethine dyes, e.g., pyrazolo[5,1-c]-1,2,4-triazoleazomethine dyes disclosed in U.S. Patent 3,725,067, pyrazolo[1,5-b]-1,2,4-triazoleazomethine dyes disclosed in U.S. Patent 4,540,654, and pyrazolo[1,5-d]tetrazoleazomethine dyes disclosed in Japanese Patent Application (OPI) No. 33552/85 give sharp visible absorption spectra as compared to conventional pyrazoloneazomethine dyes, and hence give a clearer hue as well as a lower side absorption described above.

However, recently it has been required to further improve the image quality of silver halide color photographic materials. Thus, dyes which can give sharper absorption spectra and have a clearer hue than the above-described pyrazoloazoleazomethine dyes have been desired.

Also, the molecular extinction coefficient of the pyrazoloazoleazomethine dyes described above is at most 6×10^4 $\ell.mol^{-1}.cm^{-1}$, and hence dyes having a higher molecular extinction coefficient have been desired.

DE-A-3605279 relates to a light-sensitive silver halide photographic material comprising a magenta coupler of general formula



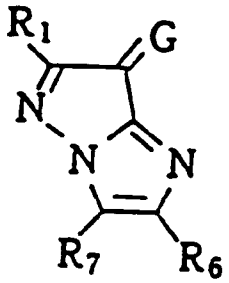
wherein R e.g. represents a sulfonyl group, a sulfinyl group or a cyano group, X represents a hydrogen atom or a group capable of being eliminated upon reaction with the oxidation product of a color developer, and Z represents a nitrogen-containing heterocyclic ring. The photographic material in accordance with DE-OS-3605279 exhibits an excellent color reproducibility and the magenta coupler(s) used are resistant to effects of light, heat and dampness.

Dyes having a higher molecular extinction coefficient can give a desired optical density with a smaller amount thereof, and hence when these dyes are utilized as, for example, dyes for forming color images of silver halide color photographic material, the thickness of emulsion layer(s) of the color photographic material can be greatly reduced, to thus reduce the cost for the silver halide in such photographic materials, and furthermore, such silver halide photographic materials provide images having an improved sharpness.

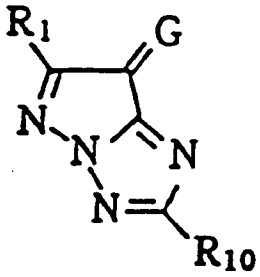
It is the object underlying the present invention to provide novel pyrazoloazoleazomethine dyes having sharp visible absorption spectra and clear color, having a high molecular extinction coefficient, providing a desired optical density in a small amount and having an adsoprtion at a longer wavelength region.

The object of the present invention is attained with a pyrazoloazoleazomethine dye represented by formulae (II), (IV), (V), (VI) or (VII),

(II)



(IV)



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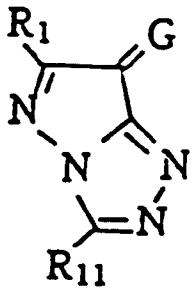
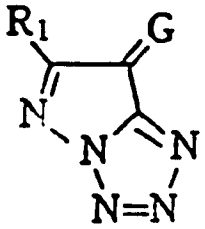
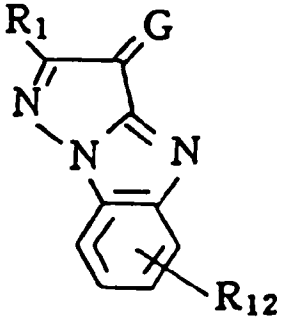


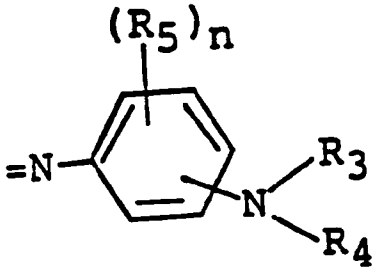
(VI)



(VII)



wherein G represents a structural moiety represented by formula (VIII)



(VIII)

wherein R_1 represents a substituent having a Hammett's substituent contstant value of at least 0.6; R_3 and R_4 each represents a hydrogen atom or a substituted or unsubstituted alkyl group and R_5 represents a hydrogen atom, a chlorine or a substituted or unsubstituted alkyl group; n represents 1 or 2;

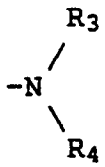
R_6, R_7, R_{10}, R_{11} and R_{12} each represents a hydrogen atom, a halogen atom, an alkyl group, an aralkyl group, an alkenyl group, an alkynyl group, a cycloalkyl group, a cycloalkenyl group, an aryl group, a heterocyclic group, a cyano group, an alkoxy group, an aryloxy group, an acylamino group, an anilino group, a ureido group, a sulfamoylamino group, an alkylthio group, an arylthio group, an alkoxycarbonylamino group, a sulfonamido group, a carbamoyl group, a sulfamoyl group, a sulfonyl group, an alkoxycarbonyl group, a heterocyclic oxy group, an acyloxy group, a carbymoyloxy group, a silyloxy group, an aryloxycarbonylamino group, an imido group, a heterocyclic thio group, a sulfinyl group, a phosphonyl group, an aryloxycarbonyl group or an acyl group; or said dye being in the form of a dimer or higher polymer by combining with each other or to a polymer through divalent group at R_1, R_6, R_7, R_{10}, R_{11} or R_{12}.

Fig. 1 is a graph showing visible absorption spectra of the pyrazoloazoleazomethine dye of this invention in Example 2 and a comparison pyrazoloazoleazomethine dye in Example 2, and

Fig. 2 is a graph showing visible absorption spectra of the pyrazoloazoleazomethine dye and cyan dyes formed from comparison phenolic cyan couplers in Example 4.

As a result of various investigations for obtaining pyrazoloazoleazomethine dyes capable of attaining the above-described object, it has been found that when a specific strong electron attractive group is introduced into a substituent of the pyrazoloazole skeleton of a pyrazoloazoleazomethine dye, the visible absorption spectrum of the dye becomes sharper and the dye provides a clearer hue as well as the molecular extinction coefficient of the dye reaches as high as 9×10^4 $\ell.mol^{-1}.cm^{-1}$, and that when a specifically strong electron attractive group is introduced into the pyrazoloazole skeleton of the dye, the maximum absorption wavelength of the absorption spectrum of the dye is greatly shifted to the deep color side and the dye becomes a blue or cyan dye having the maximum absorption wavelength of over 600 pm by suitably selecting the pyrazoloazole skeleton and the specifically strong electron attractive group, while a conventional pyrazoloazoleazomethine dye is a red, magenta, or purple dye having the maximum absorption wavelength at the region of from about 520 to about 570 nm.

When R_5 in formula (VIII)is a halogen atom or an alkyl group it is preferable that R_5 and



are at the o- and p- position, respectively, of the benzene ring from the point of view of the absorbance and hue of the dye.

In the above formulae R_1 has a Hammett's substituent constant value of at least 0.6, preferably from 0.6 to 2.0, most preferably from 0.6 to 1.0.

The compound represented by formula (VII) may have plural groups represented by R_{12}.

The pyrazoloazoleazomethine dyes of this invention are more preferably the pyrazoloazoleazomethine dyes shown by formula (IV), (V), or (VI), and further preferably those shown by formula (IV) described above.

More specificially, examples of R_6, R_7, R_{10}, R_{11} and R_{12} are a hydrogen atom, a halogen atom (e.g., a chlorine atom or a bromine atom) or a straight chain or branched alkyl group, an aralkyl group, an alkenyl group, an alkynyl group, a cycloalkyl group, and a cycloalkenyl group said groups having 1 to 32 carbon atoms, each of which may be substituted by a substituent which is connected thereto through an oxygen atom, a nitrogen atom, a sulfur atom, or a carbonyl group, a hydroxy group, an amino group, a nitro group, a carboxy group, a cyano group, or a halogen atom; and examples of these alkyl groups are a methyl group, a propyl group, a t-butyl group, a trifluoromethyl group, a tridecyl group, a 2-methanesulfonylethyl group, a 3-(3-pentadecylphenoxy) propyl group, a 3-{4-{2-[4-(4-hydroxyphenylsulfonyl) phenoxyl]-dodecaneamido}phenyl}propyl group, a 2-ethoxytridecyl group, a trifluoromethyl group, a cyclopentyl group and a 3-(2,4-di-t-amylphenoxy)-propyl group, an aryl group (e.g., a phenyl group, a 4-t-butylphenyl group, a 2,4-di-t-amylphenyl group and a 4-tetradecaneamidophenyl group), a heterocyclic group (e.g., a 5- to 7-membered heterocyclic ring having at least one of N, O and S atoms, such as a 2-furyl group, a 2-thienyl group, a 2-pyrimidinyl group and a 2-benzothiazolyl group), a cyano group, an alkoxy group (e.g., a

methoxy group, an ethoxy group, a 2-methoxyethoxy group, a 2-dodecylethoxy group and a 2-methanesulfonylethoxy group), an aryloxy group (e.g., a phenoxy group, a 2-methylphenoxy group and a 4-t-butylphenoxy group), an acylamino group (e.g., an acetamido group, a benzamido group, a tetradecaneamido group, an α-(2,4-di-t-amylphenoxy)butylamido group, a γ-(3-t-butyl-4-hydroxyphenoxy)-butylamido group and an a-{4-(4-hydroxyphenylsulfonyl)phenoxy}decaneamido group), an anilino group (e.g., a phenylamino group, a 2-chloroanilino group , a 2-chloro-5-tetradecaneamidoanilino group, a 2-chloro-5-dodecyloxycarbonylanilino group, an N-acetyl-anilino group and a 2-chloro-5-{α-(3-t-butyl-4-hydroxyphenoxy)dodecaneamido}anilino group), a ureido group (e.g., a phenylureido group, a methylureido group and an N,N-dibutylureido group), a sulfamoylamino group (e.g., an N,N-dipropylsulfamoylamino group and an N-methyl-N-decylsulfamoylamino group), an alkylthio group (e.g., a methylthio group, an octylthio group, a tetradecylthio group, a 2-phenoxyethylthio group, a 3-phenoxypropylthio group and a 3-(4-t-butylphenoxy) propylthio group), an arylthio group (e.g., a phenylthio group, a 2-butoxy-5-t-octylphenylthio group, a 3-pentadecylphenylthio group, a 2-carboxyphenylthio group and a 4-tetradecaneamidophenylthio group), an alkoxycarbonylamino group (e.g., a methoxycarbonylamino group and a tetradecyloxycarbonylamino group), a sulfonamido group (e.g., a methanesulfonamido group, a hexadecanesulfonamido group, a benzenesulfonamido group, a p-toluenesulfonamido group, an octadecanesulfonamido group and a 2-methyloxy-5-t-butylbenzenesulfonamido group), a carbamoyl group (e.g., an N-ethylcarbamoyl group, an N,N-dibutylcarbamoyl group, an N-(2-dodecyloxyethyl)carbamoyl group, an N-methyl-N-dodecylcarbamoyl group and an N-{3-(2,4-di-t-amylphenoxy)propyl}carbamoyl group), a sulfamoyl group (e.g., an N-ethylsulfamoyl group, an N,N-dipropylsulfamoyl group, an N-(2-dodecyloxyethyl) sulfamoyl group, an N-ethyl-N-dodecylsulfamoyl group and an N, N-diethylsulfamoyl group), a sulfonyl group (e.g., a methanesulfonyl group, an octanesulfonyl group, a benzenesulfonyl group and a toluenesulfonyl group), an alkoxycarbonyl group (e.g., a methoxycarbonyl group, a butyloxycarbonyl group, a dodecyloxycarbonyl group and an octadecyloxycarbonyl group), a heterocyclic oxy group (wherein the heterocyclic group preferably is a 5-to 7- membered heterocyclic ring having at least one of N, O and S atoms, such as e.g. a 1-phenyltetrazole-5-oxy group and a 2-tetrahydopyranyloxy group), an acyloxy group (e.g., an acetoxy group), an acylaminoxy group (e.g., an acetylaminoxy group and a benzoylaminoxy group), a silyloxy group (e.g., a trimethylsilyloxy group and a dibutylmethylsilyloxy group),an aryloxycarbonylamino group (e.g., a phenoxycarbonylamino group), an imido group (e.g., an N-succinimido group, an N-phthalimido group and a 3-octadecenylsuccinimido group), a heterocyclic thio group (wherein the heterocyclic group preferably is a 5- to 7- membered heterocyclic ring having at least one of N, O and S atoms, such as e.g., a 2-benzothiazolylthio group, a 2,4-di-phenoxy-1,3,5-triazole-6-thio group and a 2-pyridylthio group), a sulfinyl group (e.g., a dodecanesulfinyl group, a 3-pentadecylphenylsulfinyl group and a 3-phenoxypropylthio group), a phosphonyl group (e.g., a phenoxyphosphonyl group, an octyloxyphosphonyl group, a phenyl phosphonyl group), an aryloxycarbonyl group (e.g., a phenoxycarbonyl group), an acyl group (e.g., an acetyl group, a 3-phenylpropanoyl group, a benzoyl group and a 4-dodecyloxybenzoyl group.

In addition, when R_1, R_6, R_7, R_{10}, R_{11}, and R_{12} in formulae (II) to (VII) described above and R_3, R_4, and R_5 in formula (VIII) described above are an alkyl group or an alkyl-containing group (e.g., an alkoxy group, an alkylthio group, an alkoxycarbamoyl group or an alkoxycarbonyl group), the carbon atom number of the alkyl group is usually from 1 to 50, and preferably from 1 to 40, and more preferably from 1 to 32, and when these groups an acyl group, the carbon atom number thereof is same as above.

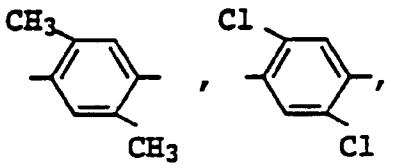
Particular examples of groups having a Hammett's substituent constant σ_p value of at least 0.6 are a cyano group, a nitro group, a trialkylammonium group (e.g., a trimethylammonium group, a triethylammonium group, a tributylammonium group, a trioctylammonium group and a tridecylammonium group), a triarylammonium group (e.g., a triphenylammonium group and a tritolylammonium group), a dialkylsulfonium group (e.g., a dimethylsulfonium group and a diethylsulfonium group), a diarylsulfonium group (e.g., a diphenylsulfonium group), a perfluoroalkylsulfinyl group (e.g., a trifluoromethylsulfinyl group, a pentfluoroethylsulfinyl group, a heptafluoropropylsulfinyl group and a perfluorooctylsulfinyl group), an ω-hydroperfluoroalkylsulfinyl group (e.g., an ω-hydroperfluorooctyl-sulfinyl group and an ω-hydroperfluorododecylsulfinyl group), an alkanelsulfonyl group (e.g., methanesulfonyl group, difluoromethanesulfonyl group, a trifluoromethanesulfonyl group, a dichloromethanesulfonyl group, an ethanesulfonyl group, a propanesulfonyl group, an octanesulfonyl group, a decanesulfonyl group, a pentafluoroethanesulfonyl group, a heptafluoropropanesulfonyl group, a perfluorooctanesulfonyl group and an ω-hydroperfluorooctanesulfonyl group), an arylsulfonyl group (e.g., a phenylsulfonyl group, a tolylsulfonyl, group and a pentafluorophenylsulfonyl group), a β-carboxyvinyl group and a β,β-dicyanovinyl group. These groups may be those represented by R_1, R_6, R_7, R_{10}, R_{11}, and R_{12}.

These examples are further described, for example, in C. Hansch et al., Substituent Constants For Correlation Analysis in Chemistry and Biology, John Wiley & Sons, New York (1979), C. Hansch et al.,

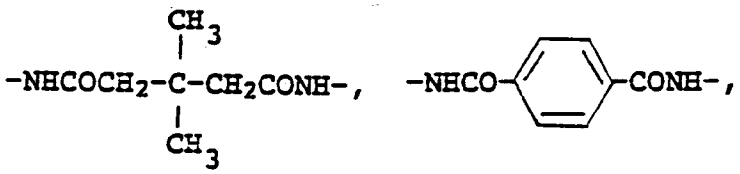
Journal of Medicinal Chemistry, Vol. 16, pp. 1207-1216 (1973) and C. Hansch et al., ibid, Vol. 20, pp. 304-306 (1977).

However, it is not unusual that different values are reported by different reporters for the same substituent, and in that case it is preferred to use values described in the last two references (both, C. Hansch et al) listed above. Further with respect to substitutuents which are not discribed in the above references, the measurement can be performed in accordance with the definition described in L.P. Hammett, Physical Organic Chemistry, (McGraw-Hill, 1970).

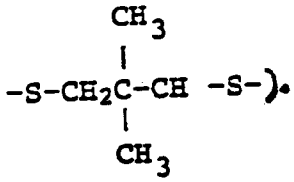
Furthermore, R_1, R_6, R_7, R_{10}, R_{11}, or R_{12} in formulae (II) to (VII) can be divalent groups forming a bis-compound. Examples of such a divalent group are a substituted or unsubstituted alkylene group (the alkylene group includes group having at least one oxygene atom in the hydrocarbon chain; e.g., a methylene group, an ethylene group, a 1,10-decylene group and $-CH_2CH_2-O-CH_2CH_2-$,), a substituted or unsubstituted phenylene group (e.g., a 1,4-phenylene group, a 1,3-phenylene group,



etc.), -NHCO-L_1-CONH (wherein L_1 represents a substituted or unsubstituted alkylene or phenylene group, e.g., $-NHCOCH_2CH_2CONH$,

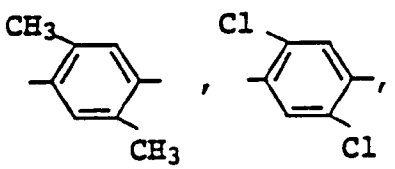


etc.), -S-L_2-S-group (wherein L_2 represents a substituted or unsubstituted alkylene group, e.g., $-S-CH_2CH_2-S-$,

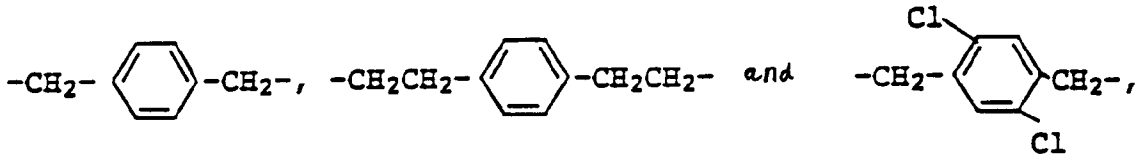


The pyrazoloazoleazomethine dye of this invention represented by formulae (II) to (VII) described above may be a polymer bonded to a polymer main chain through R_1, R_6, R_7, R_{10}, R_{11}, or R_{12} as a divalent linkage group.

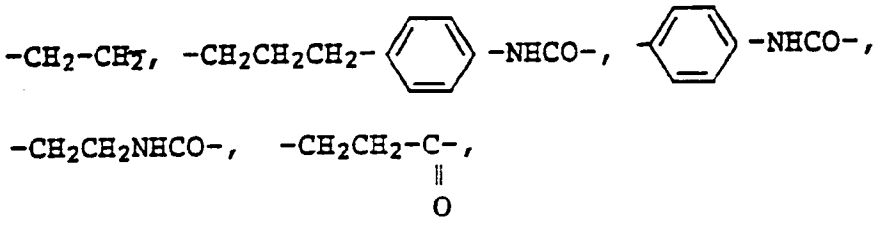
Such a divalent linkage group includes a group formed by combining the groups selected from an alkylene group (substituted or unsubstituted alkylene group, such as a methylene group, an ethylene group, a 1,10-decylene group and $-CH_2CH_2OCH_2CH_2-$), a phenylene group (substituted or unsubstituted phenylene group, such as a 1,4-phenylene group, a 1,3-phenylene group,



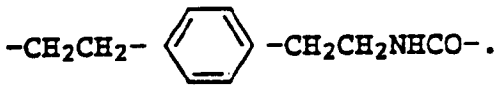
etc.), -NHCO-, -COHN-, -O-, -OCO-, and aralkylene group (e.g.)



Preferred examples of the linkage group are-NHCO-,



-CONH-CH_2CH_2NHCO-,
-CH_2CH_2O-CH_2CH_2-NHCO-, and



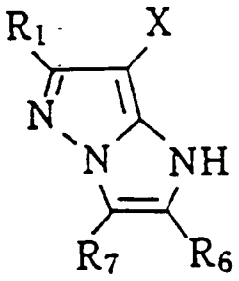
When the pyrazoloazoleazomethine dye represented by formulae (II) to (VII) described above is a polymer, it may be a copolymer with an ethylenically unsaturated monomer such as e.g., styrene, methyl acrylate, ethyl acrylate, butyl methacrylate, acrylic acid or acrylamide.

The non-coloring ethylenically unsaturated monomer for forming a copolymer with the solid water-insoluble monomer, pyrazoloazole can be selected so that the copolymer formed has desired physical and/or chemical properties such as e.g., solubility, compatibility with a binder for the dye composition, flexibility and heat-stability.

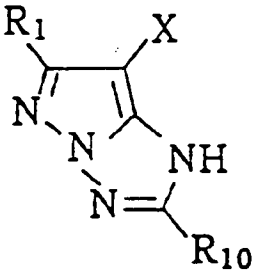
The polymeric pyrazoloazoleazomethine dye of this invention may be water-soluble or water-insoluble, but is particularly preferably a polymer latex.

Preferred synthesis intermediates for producing the pyrazoloazoleazomethine dyes of this invention shown by formulae (II) to (VII) described above are pyrazoloazoles represented by following formulae (IIA) to (VIIA):

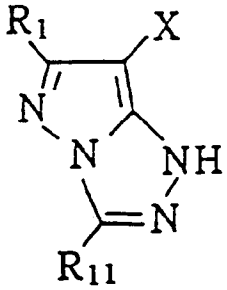
(IIA)



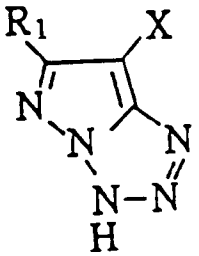
(IVA)



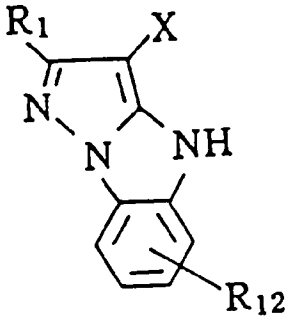
(VA)



(VIA)



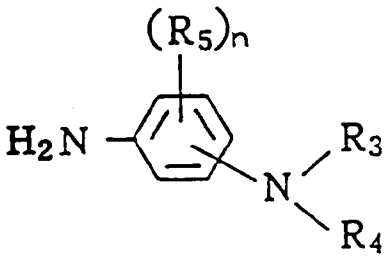
(VIIA)



In formulae (IIA) to (VIIA), R_1 and R_6 to R_{12} have the same meanings as defined above and X represents a hydrogen atom, a halogen atom, or a group capable of being released by a coupling reaction with the oxidation product of an aromatic primary amino compound. (With respect to such a group capable of being released, detailed descriptions can be seen in, for example, U.S. Patent 4,540,654, column 4, line 30 to column 5, line 24.)

These synthesis intermediates as described above can be produced by the methods disclosed, for example, in U.S. Patents 3,061,432, 3,725,067, 4,500,630, and 4,540,654, Japanese Patent Application (OPI) Nos. 33552/85, 43659/85, and 186567/85.

The pyrazoloazoleazomethine dyes of this invention represented by formulae (II) to (VII) can be produced by oxidative-coupling the synthesis intermediates shown by formulae (IIA) to (VIIA) and an aromatic primary amine represented by formula (VIIIA)



(VIIIA)

wherein R_3 and R_4, which may be the same or different, each represents a hydrogen atom or an alkyl group, which may have a substituent; R_5 represents a hydrogen atom, a halogen atom, or an alkyl group, which may have a substituent; and n represents a number of substituents of R_5 which is 1 or 2; when n is 2, said R_5 groups may be the same or different.

The amine represented by formula (VIIIA) described above can be used as a salt with a mineral acid or an organic acid and in this case, the air oxidation can be easily prevented and the dissolution rate of the compound can be improved.

In formula (VIIIA), R_3 and R_4 preferably represent a hydrogen atom, an alkyl group or a substituted alkyl group such as a hydroxyalkyl group, an alkoxyalkyl group, an alkoxyalkoxyalkyl group, and an alkylsulfonamidoalkyl group, and R_5 may represent an alkyl group substituted with an alkoxy group or a halogen atom.

Furthermore, examples of the alkyl group in the phenylenediamine derivatives shown by formula (VIIIA) described above and the alkyl moiety of the alkoxy group and the substituted alkyl group are lower alkyl groups having from 1 to 4 carbon atoms, such as e.g., methyl, ethyl, n-propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl and t-butyl, and higher alkyl groups having from 5 to 18 carbon atoms, such as e.g., n-amyl, d1-2-methyl-1-butyl, iso-amyl, sec-amyl, t-amyl, n-hexyl, methylamyl, 2-ethyl-butyl, n-heptyl, 2-heptyl, 3-heptyl, n-octyl, 2-octyl, 2-ethylhexyl, n-dodecyl, n-octadecyl and cyclohexyl, which include straight chain, branched or cyclic alkyl groups.

Examples of a halogen atom are a chlorine atom, a bromine atom, and an iodine atom.

The aromatic primary amines shown by formula (VIIIA), which are used for producing the pyrazoloazoleazomethine dyes of this invention are preferably ortho- or para-phenylenediamines, and more preferably para-phenylene-diamines. Specific examples thereof are those having N-alkyl group, such as for example,

D 1) 4-amino-N-ethylaniline,
D 2) 4-amino-N,N-diethylaniline and
D 3) 4-amino-3-methyl-N,N-diethylaniline; those having N-hydroxyalkyl group such as, for example,
D 4) 4-amino-N-ethyl-N-(β-hydroxyethyl)aniline and
D 5) 4-amino-3-methyl-N-ethyl-N-(β-hydroxyethyl)aniline;

those having N-alkoxyalkyl group such as, for example,

D 6) 4-amino-3-methyl-N-ethyl-(β-methoxyethyl)aniline,
D 7) 4-amino-3-methyl-N-ethyl-N-methoxybutylaniline,
D 8) 4-amino-3-methyl-N-ethyl-N-(β-ethoxyethyl)aniline,
D 9) 4-amino-3-propyl-N-ethyl-N-(β-methoxyethyl)aniline,
D 10) 4-amino-3-propyl-N-ethyl-N-(β-methoxyethyl)aniline,
D 11) 4-amino-3-methoxy-N-ethyl-N-(β-methoxyethyl)aniline and
D 12) 4-amino-3-methyl-N-ethyl-N-(β-butoxyethyl)aniline;

those having N-alkoxyalkoxyalkyl group such as, for example,

D 13) 4-amino-3-ethyl-N-ethyl-N-[β-(β-methoxyethoxy)ethyl]aniline,
D 14) 4-amino-3-methyl-N-ethyl-N-[β-(β-ethoxyethoxy)ethyl]aniline,
D 15) 4-amino-3-methyl-N-ethyl-N-[β-(β-butoxyethoxy)ethyl]aniline,
D 16) 4-amino-3-methyl-N-methyl-N-[β-(β-ethoxyethoxy)ethyl]aniline,
D 17) 4-amino-N-ethy1-N-[β-(β-methoxyethoxy)ethyl]aniline and
D 18) 4-amino-N-ethy1-N-[β-(β-ethoxyethoxy)ethyl]aniline; and

those having N-alkylsulfonamidoalkyl group such as, for example,

D 19) 4-amino-N-ethy1-N-(β-methanesulfonamidoethyl)aniline,
D 20) 4-amino-3-methyl-N-ethyl-N-(β-methanesulfonamidoethyl)aniline,
D 21) 4-amino-3-chloro-N-ethyl-N-(β-methanesulfonamidoethyl)aniline and
D 22) 4 -amino-N-ethyl-N-(β-methylsulfonamidoethyl)-3,5-xylidine.

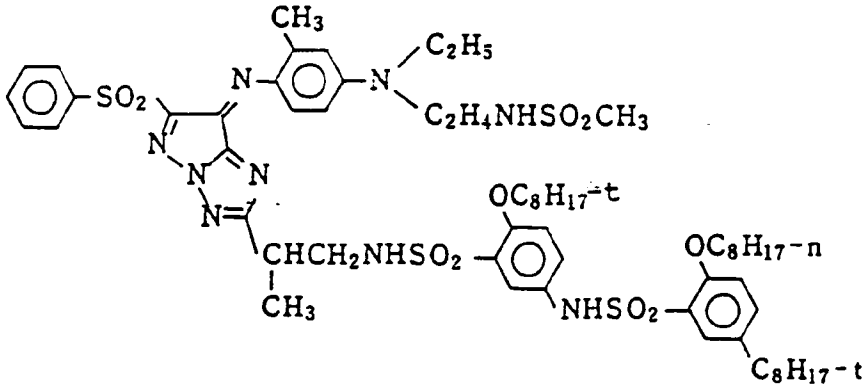
Examples of the salts of the phenylenediamine derivatives, are inorganic acid salts, i.e., mineral acid salts such as e.g., hydrochlorides, sulfates, nitrates, phosphates, carbonates, and hydrohalogenic acid salts such as hydrochlorides, hydrobromides and hydroiodides, and organic acid salts such as e.g., aliphatic carboxylates such as formates, acetates and propionates, aromatic carboxylates such as e.g., benzoate naphthalene-α-carboxylates and naphthalene-β-carboxylates, aliphatic sulfonates such as e.g., methanesulfonates, and aromatic sulfonates such as e.g., naphthalene-α-sulfonates, naphthalene-β-sulfonates and p-toluenesulfonates.

They may be properly selected according to the production conditions for the dyes. For example, when the salts are used as photographic color developing agents, it is preferred to use salts having no adverse influences on the photographic properties. For this purpose, the phenylenediamine derivatives are usually used as the mineral acid salts such as sulfates or aromatic sulfonates such e.g., as p-toluenesulfonates.

As the phenylenediamines for use in this invention, the above-described compounds D 3), D 5), D 6), D 19) and D 20) are particularly preferred from the view point of providing a particularly good hue. Useful for controlling the coupling speed is a substituent at the 3-position which is an electron attractive group such as e.g., a chlorine atom and functions to increase the coupling speed, and which is an electron donative substituent such as e.g., a methyl group and functions to delay the coupling speed.

Preferred examples of the pyrazoloazoleazomethine dyes of this invention are illustrated below.

Compound 1



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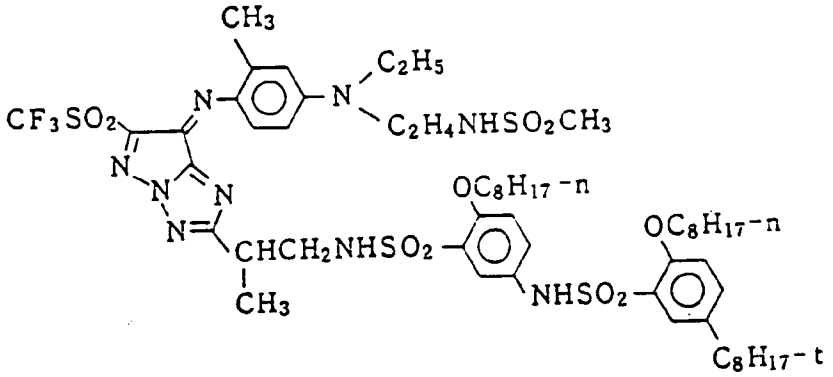

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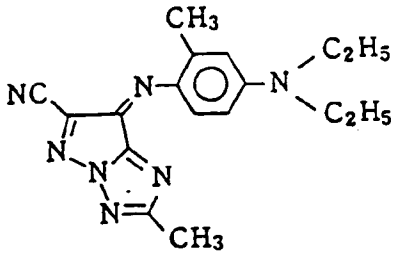

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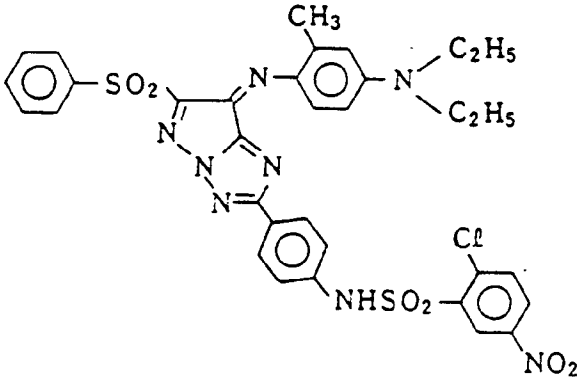

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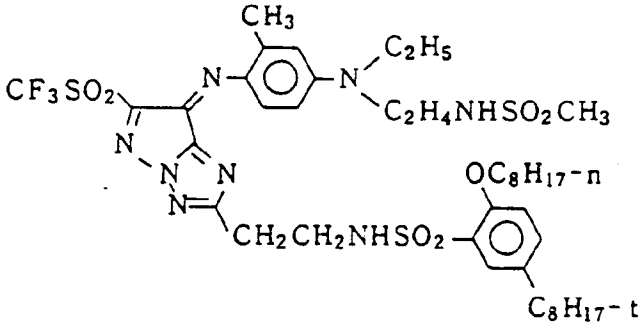

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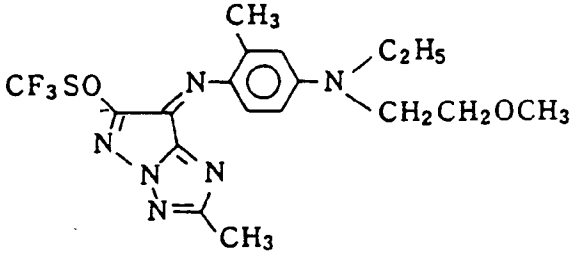
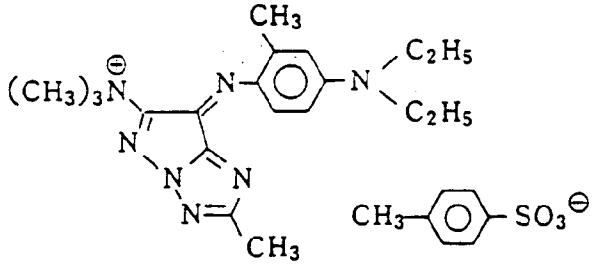
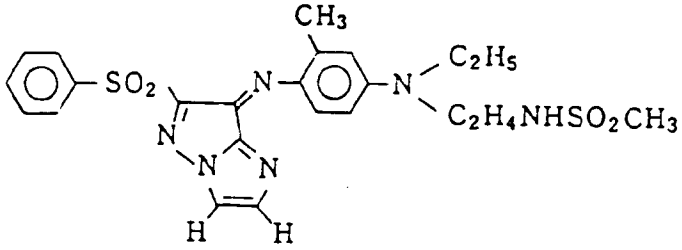


Compound 8



Compound 9



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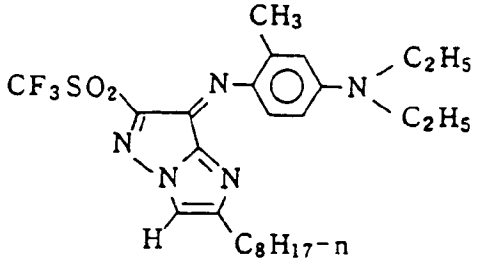

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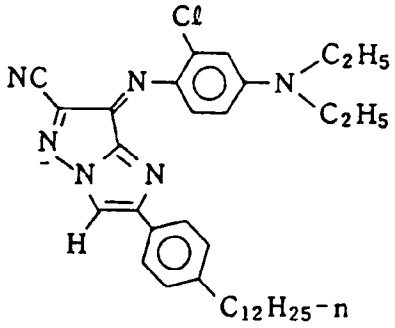

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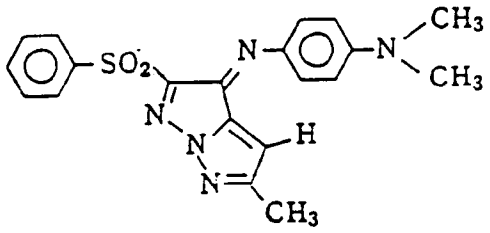

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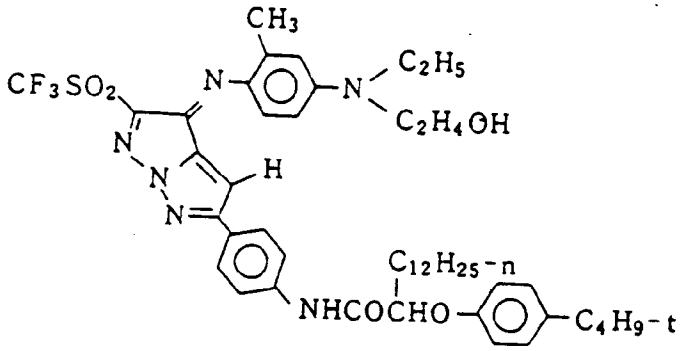

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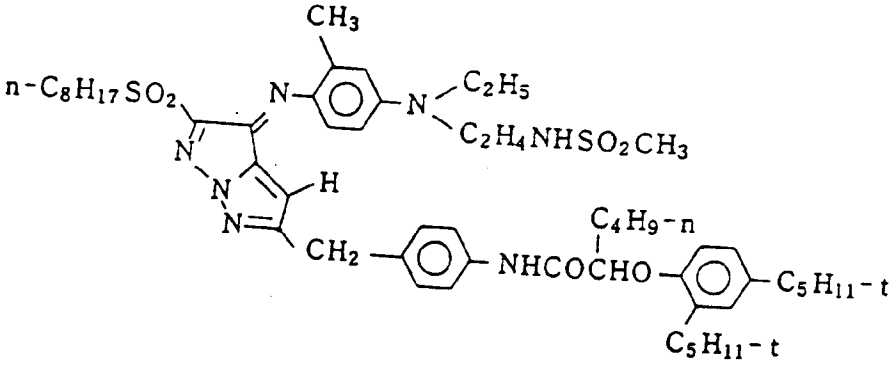

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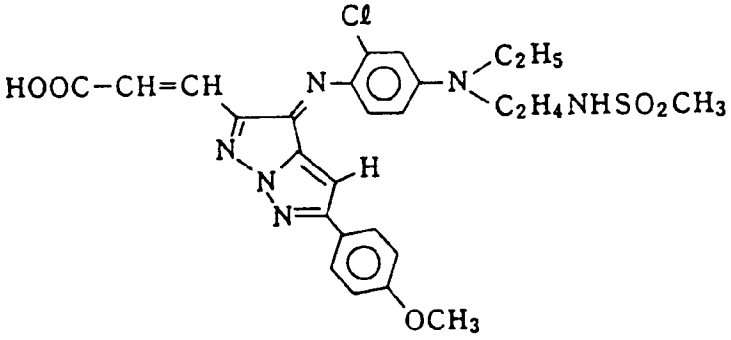

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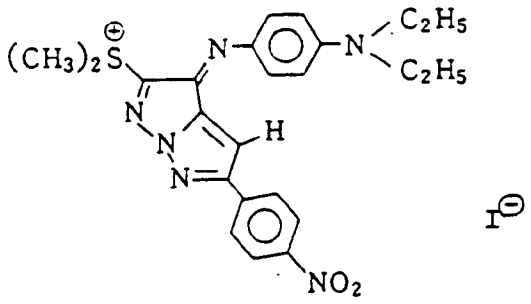

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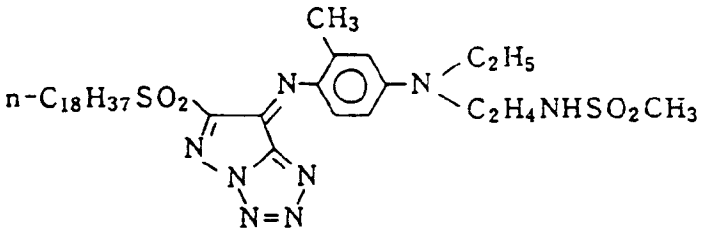

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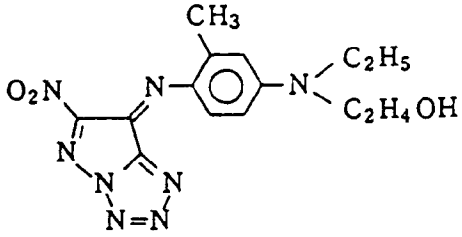

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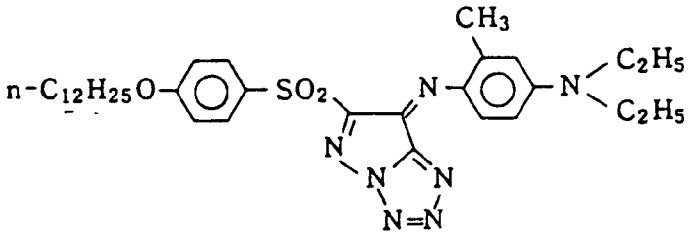

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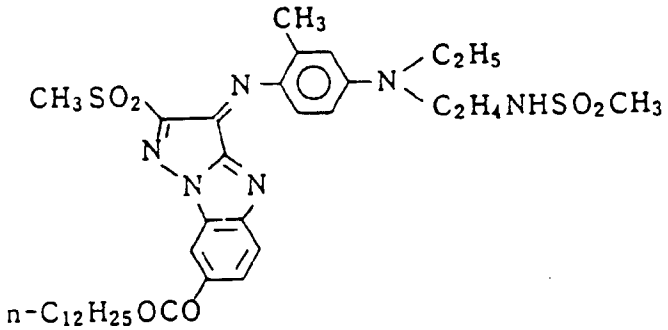

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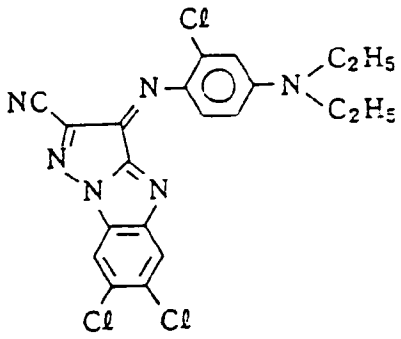

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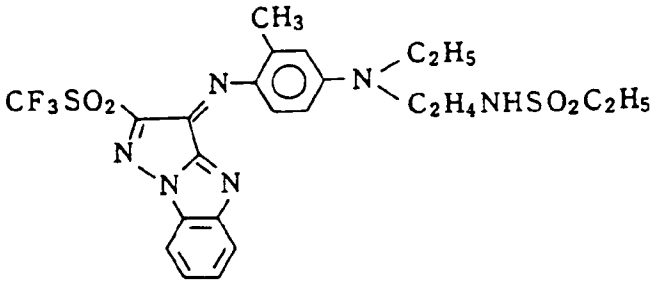

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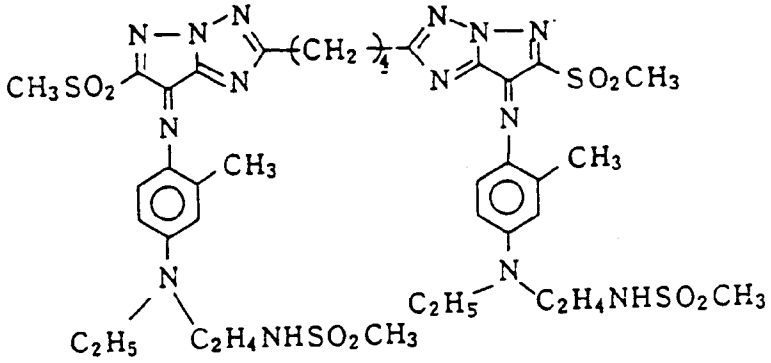

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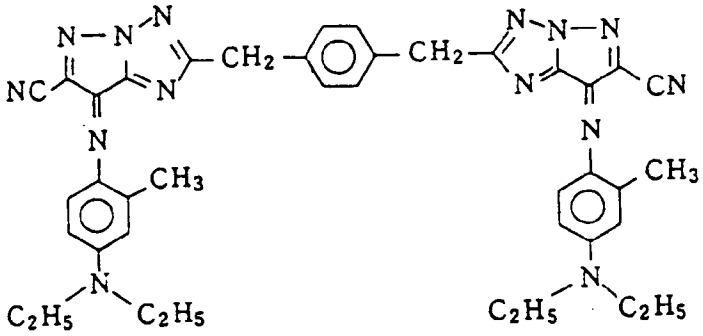
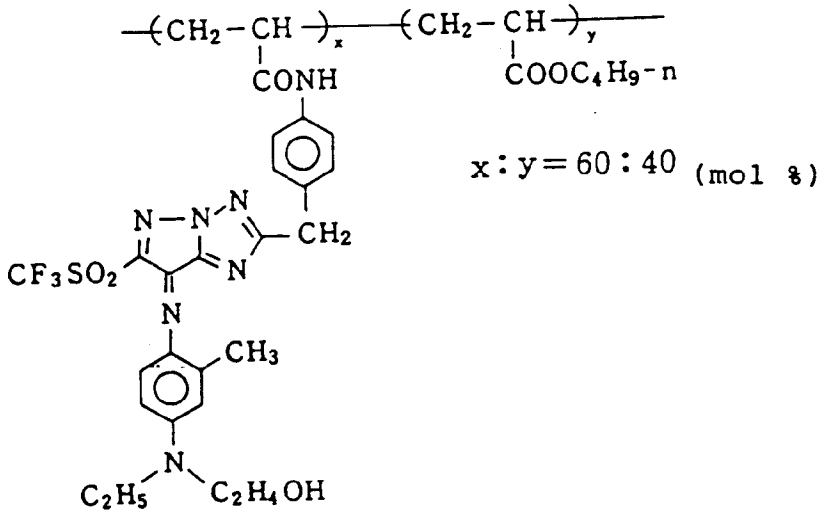


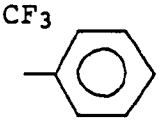
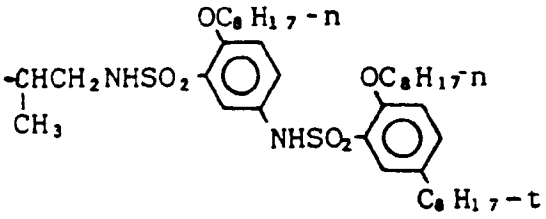
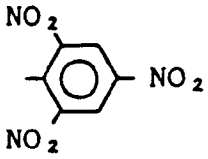
Compound 26



x : y = 60 : 40 (mol %)

The Hammett's substituent constant σ_p values of the substituents in the above-described compounds are shown in Table 1 below.

Table 1

Group	σ_p
- SO_2CH_3	0.72
- NO_2	0.78
- SO_2CF_3	0.93
- SO_2 	0.70
	-0.15
- CN	0.66
- CF_3	0.54
	0.30
- $SOCF_3$	0.69
- $N^{\oplus}(CH_3)_3$	0.82
- CH_3	- 0.17
- CH = CH - COOH	0.90
- $S^{\oplus}(CH_3)_2$	0.90

When the σ_p value of the aforesaid substituent R_1 in the pyrazoloazoleazomethine dyes of this invention is at least 0.6 and larger, the dyes have the tendency of shifting to the deep color side. The shifting extent of Δ_{max} depends on the kind of the pyrazoloazole skeleton (formula (II) to (VII)) and on the σ_p value of the substituent, but is generally in the range of from about 50 n.m. to 100 n.m.

The pyrazoloazoleazomethine dyes shown by formulae (II) to (VII) can be synthesized in the co-existence of the pyrazoloazole couplers shown by formulae (IIA) to (VIIA) described above, the

phenylenediamine shown by formula (VIIIA) described above, and an oxidizing agent. It is considered that in the coupling reaction, the azomethine dye is formed after forming a leuco dye by a nucleophilic attack of a coupler anion to a quinonediimine formed by the oxidation of the phenylenediamine as described in T.H. James, The Theory of the Photographic Process, 4th edition,Chapter 12, (1977, Macmillan). The reaction preferably proceeds under basic conditions and the reaction medium may be an organic solvent, an aqueous organic solvent, or an aqueous solution. When the reaction proceeds in a basic aqueous solution, the coupler may be an oil drop-in water dispersion and further the dispersion may exist in a hydrophilic colloid medium such as gelatin.

As the oxidizing agent, an organic or inorganic oxidizing agent having an electric potential capable of oxidizing the phenylenediamine can be used and the oxidizing agent may be dissolved in a reaction medium or may be dispersed therein.

When X in formulae (II) to (VII) described above is a hydrogen atom, from 0.1 to 10 mols, and preferably from 0.5 to 2 mols of the phenylenediamine represented by formula (VIIIA) is used to one mole of the coupler represented by formula (IIA) to (VIIA) and at least 4 equivalents, and preferably 4.4 to 20 equivalents of oxidizing agent is used. When X is not a hydrogen atom, the dye of formulae (II) to (VII) can be synthesized by the same manner as above except that the amount of the oxidizing agent is at least 2 equivalents, and preferably 2.2 to 10 equivalents.

When the reaction is performed in an aqueous medium, the pH is higher than 8, and preferably the coupling reaction is performed in the pH range of from 10 to 12.

As the oxidizing agent, silver halide, hydrogen peroxide, manganese dioxide, potassium persulfate, oxygen, and other compounds described in Fieser and Fieser, Organic Reagents, can be used.

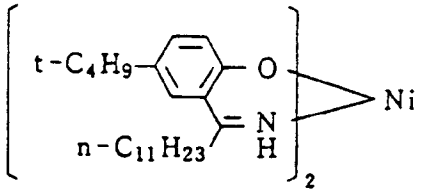
The dyes of this invention can be imagewise formed according to the process of silver halide color photography as described in U.S. Patent 4,540,654.

For improving the stability of the dyes of this invention to light or heat, known stabilizers may be used together with the dyes. Examples of organic compounds capable of improving the stability of the dyes are hydroquinone derivatives described in U.S. Patents 3,935,016 and 3,982,944, hydroquinone diether derivatives described in U.S. Patent 4,254,216 and Japanese Patent Application (OPI) No. 21004/80, phenol derivatives described in Japanese Patent Application (OPI) No. 145530/79, spiroindane derivatives and methylenedioxybenzene derivatives described in British Patent Application (OPI) Nos. 2,077,455 and 2,062,888, and Japanese Patent Application (OPI) No. 90155/86, chroman derivatives, spirochroman derivatives, and coumaran derivatives described in U.S. Patents 3,764,337, 3,432,300, 3,574,627, and 3,573,050, Japanese Patent Application (OPI) Nos. 152225/77, 20327/78, 17729/78, and 90156/86, hydroquinone monoether derivatives and para-aminophenol derivatives described in Japanese Patent Application (OPI) No. 6321/80, British Patent 1,347,556, British Patent Application (OPI) No. 2,066,975, and Japanese Patent Publication No. 12337/79, and bisphenol derivatives described in Japanese Patent Publication No. 31625/73 and U.S. Patent 3,700,455.

For improving the stability of the dyes to light or heat, the use of a metal complex together with the dye is also effective. Examples of such a metal complex are described in U.S. Patent 4,245,018 and Japanese Patent Application (OPI) No. 97353/85.

Specific examples of the metal complex are illustrated below.

M－1)



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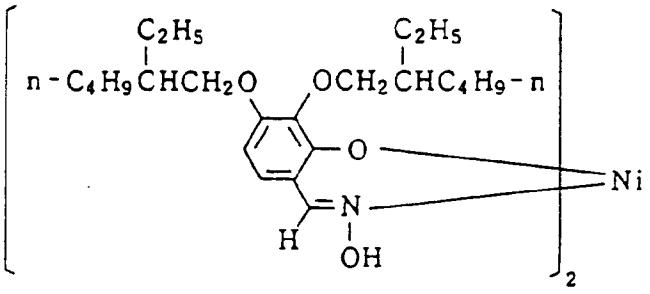

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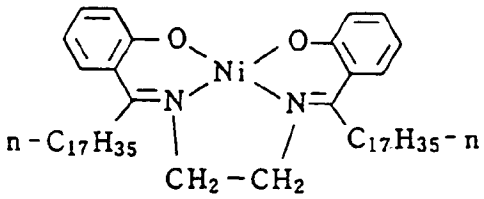

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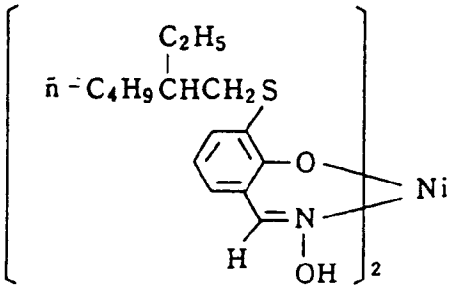
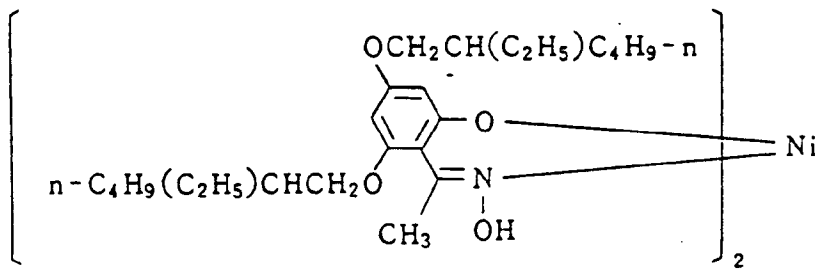


M－5)



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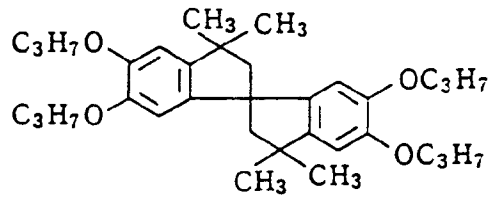

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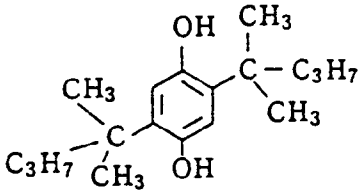

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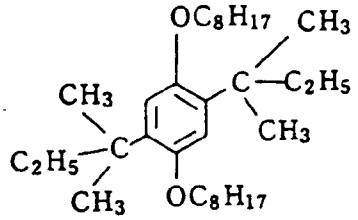

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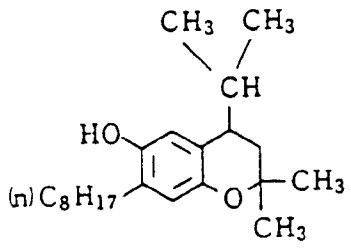

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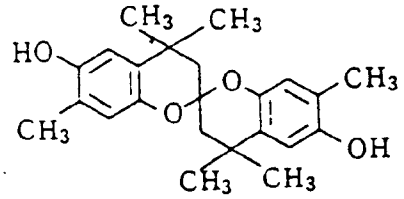

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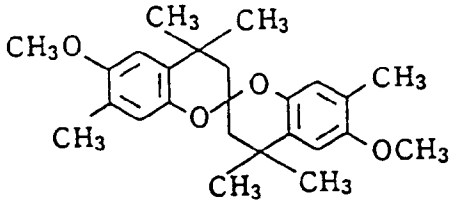

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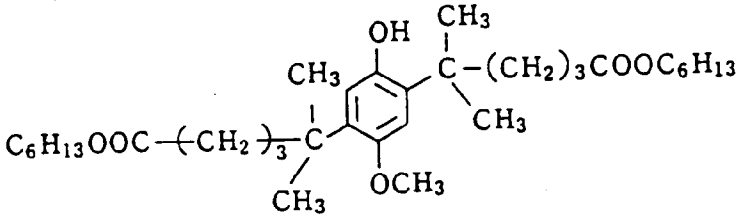

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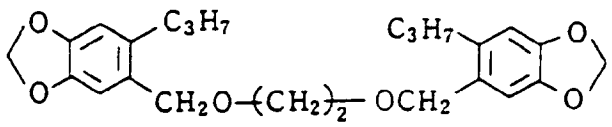

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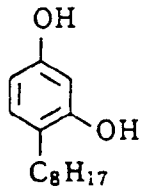

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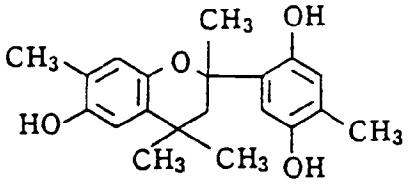

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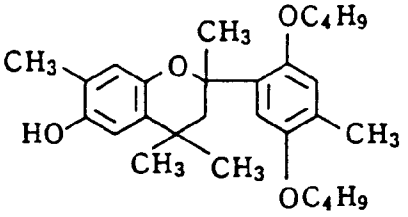

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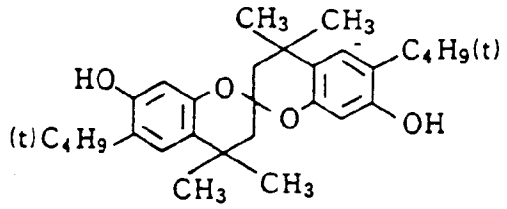

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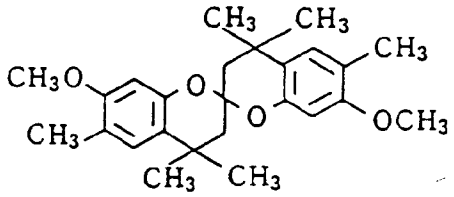

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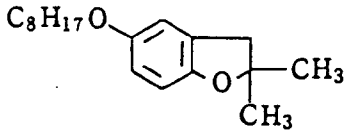

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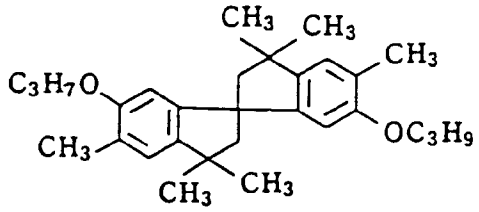


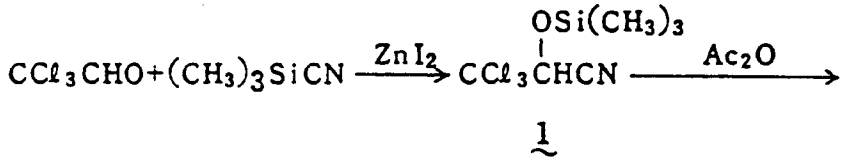
The pyrazoloazoleazomethine dyes of this invention are useful as magenta dyes or cyan dyes for forming color images. The dyes of this invention are also useful as magenta filter dyes or cyan filter dyes for silver halide color photographic materials and also useful as irradiation preventing dyes and antihalation dyes. Furthermore, the dyes of this invention are useful as magenta filter dyes or cyan filter dyes for solid pickup tubes.

The pyrazoloazoleazomethine dyes of this invention have far sharper visible absorption spectra than conventional pyrazoloazoleazomethine dyes and hence given a very clear hue. The dyes of this invention also have a high molecular extinction coefficient and can give a desired optical density in a smaller amount.

The following examples serve to illustrate the present invention.

Example 1 Synthesis of Compound 4

Synthesis route



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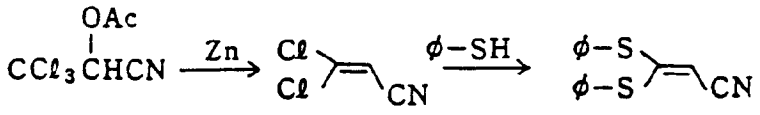

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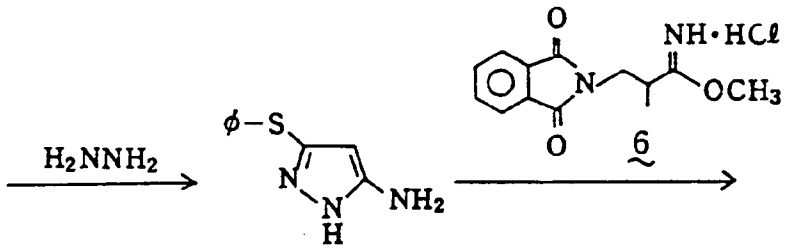

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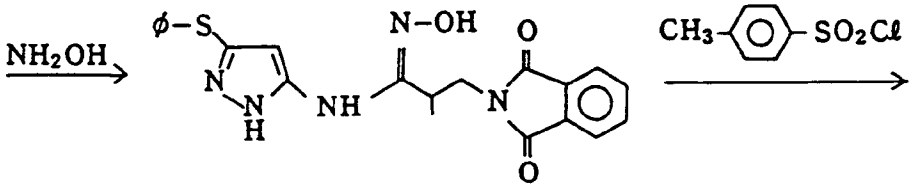

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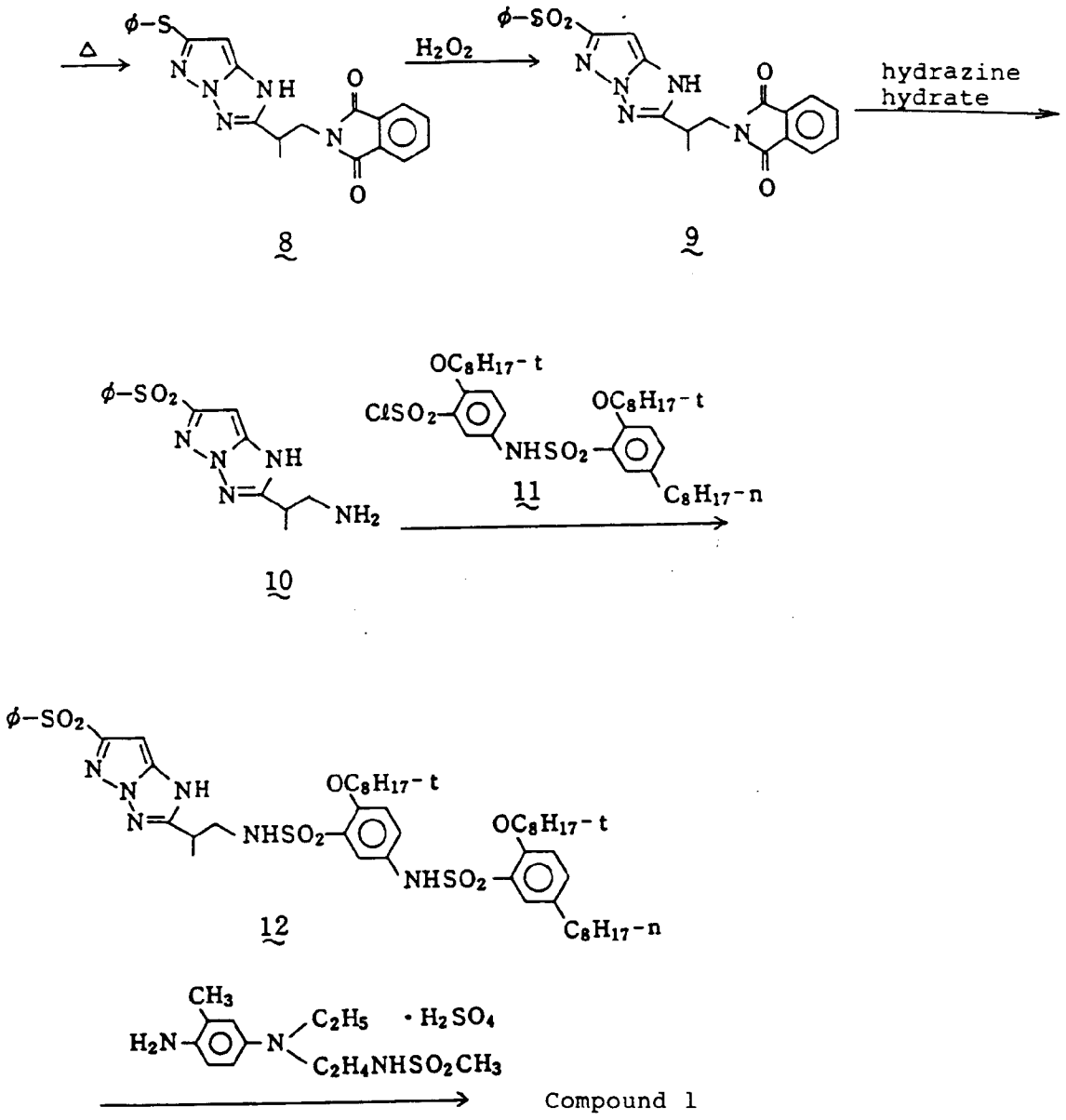


Synthesis of Intermediate 1

In a 2ℓ three neck flask equipped with a dropping funnel and a stirrer were placed 116 g (0.79 mol) of chloral (reagent grade), 24 g (0.075 mol) of zinc iodide, and 1 l of methylene dichloride, and the mixture was stirred at room temperature. Then, the mixture was cooled to a temperature of from 8 °C to 10 °C and 105 ml (0.75 mol) of methylsilylnitrile (reagent grade) were added dropwise to the mixture with stirring over a period of 30 min. The color of the mixture gradually changed from pink to brown. Thereafter, after further stirring the mixture for 3 h at room temperature, insoluble matters were removed by filtration and the filtrate formed was distilled under reduced pressure. The ^{1}H NMR spectra (solvent: CCl_4) of the liquid obtained showed δ = 0.50 ppm (belonging to Si $(CH_3)_3$ of Intermediate 1), δ = 5.0 ppm (belonging to methine of Intermediate 1), and δ = 5.4 ppm (belonging to methylene dichloride). From the NMR integral strength ratio, the content of Intermediate 1 was 69.1% by weight, the amount thereof was 173 g and the yield was 93%. (δ represents a chemical shift value.)

Synthesis of Intermediate 2

In a 1ℓ three neck flask equipped with a reflux condensor and a stirrer were placed 170 g (0.69 mol) of Intermediate 1 containing 76 g of methylene dichloride and after adding thereto 354 ml of acetic anhydride, the resultant mixture was heated to 125 to 130°C for 9 h with stirring. After cooling the mixture to room temperature, insoluble matters were filtered off, and the filtrate was distilled under reduced pressure. The residue thus formed was subjected to column chromatography using a silica gel as a solid phase and chloroform as a moving phase, and the filtrate obtained was distilled under reduced pressure. Then, volatile matters were further distilled off at 100°C using a vacuum pump. The ^{1}H-NMR spectra (solvent: CCl_4) of the liquid product thus obtained showed δ = 2.1 ppm (belonging to acetic anhydride), δ = 2.3 ppm (belonging to -OAc of intermediate 2), and δ = 6.0 ppm (belonging to methine of Intermediate 2). From the integral strength ratio of the NMR, the content of Intermediate 2 as 95.6% by weight, the amount thereof was 98.1 g, and yield was 66%.

Synthesis of Intermediate 3

In a 2ℓ three neck flask equipped with a reflux condenser and a stirrer were placed 98.1 g (0.45 mol) of Intermediate 2 containing 4.5 g of acetic anhydride and after adding thereto 400 ml of tetrahydrofuran, the mixture was heat-refluxed with stirring. Then, 32.5 g of zinc powder previously treated with hydrochloric acid was gradually added to the mixture with care. Thereafter, the mixture was heat-refluxed with stirring for 30 min and then distilled under reduced pressure. After removing 140 ml of the initial fraction, the residue was distilled at one stroke. The ^{1}H NMR spectra (solvent: CCl_4) of the liquid product thus obtained showed δ = 1.8 ppm (belonging to tetrahydrofuran), δ = 3.6 ppm (belonging to tetrahydrofuran), and δ = 6.1 ppm (belonging to intermediate 3). From the integral strength ratio of the NMR, the content of Intermediate 3 was 20.9% by weight, the amount thereof was 36.6 g and the yield was 66%.

Synthesis of Intermediate 4

In a 1ℓ three neck flask equipped with a stirrer were placed 25.8 g of sodium hydroxide and 150 ml of distilled water followed by stirring to dissolve sodium hydroxide. Then 69.6 g (0.60 mol) of thiophenol was dissolved in the solution with stirring and the solution obtained was ice-cooled, whereby the solution separated into two layers with white turbidity. While stirring the solution under ice-cooling 36.6 g (0.30 mol) of Intermediate 3 containing 138.5 g of tetrahydrofuran were added dropwise to the solution over a period of 30 min. Thereafter, the mixture was stirred for 6 h at room temperature and allowed to stand overnight, whereby the solution separated into two layers. The lower layer was removed, the upper layer was placed in a separation funnel, and after adding thereto ethyl acetate and washing with an saturated aqueous solution of sodium chloride, the solution in the funnel was dried by Glauber's salt and distilled under reduced pressure. The residue was applied to column chromatogrpahy using a silica gel as fixed phase and a mixture of ethyl acetate and n-hexane (1/10 by volume: the same hereinafter) as moving phase to provide 77.0 g (yield 95%) of oily Intermediate 4. The ^{1}H-NMR spectra (solvent: CCl_4) of the product showed δ = 7.3 ppm.

Synthesis of Intermediate 5

In a 500 mℓ three neck flask equipped with a reflux condenser and a stirrer were placed 77.0 g (0.285 mol) of Intermediate 4 and 173 ml of hydrazine hydrate and then the mixture was refluxed (inside temperature 110 to 115°C) with heating - refluxing and stirring for 4 h. In this case, the system was initially in two layers, but became one layer after reaction. After allowing to stand overnight, the reaction mixture was placed in a separation funnel, added with ethyl acetate, washed with a saturated aqueous sodium chloride solution containing 0.3N sodium hydroxide, dried by Glauber's salt, and then purified by column chromatography using silica gel as fixed phase and a mixture of methanol and chloroform (1/5) as moving phase. Thus, 21.1 g (yield 39%) of light yellow oily Intermediate 5 were obtained.

The ^{1}H-NMR spectra (Solvent: $(CD_3)_2CO$, D: heavy hydrogenated acetone) of the product showed δ = 3.3 ppm, δ = 5.6 ppm, δ = 5.9 ppm, and δ = 7.2 ppm.

Synthesis of Intermediate 7

In a 300 mℓ three neck flask equipped with a reflux condenser and a stirrer were placed 21.1 g (0.11 mol) of Intermediate 5, 200 ml of methanol and 37.3 g (0.13 mol) of Intermediate 6 prepared according to the method described in Japanese Patent Application (OPI) No. 171956/84 and then the mixture was stirred for 4 h at room temperature. In this case, it was confirmed by thin layer chromatography that Intermediate 5 had vanished. Thereafter, a solution of 115 g of hydroxylamine hydrochloride and 13.5 g of sodium acetate dissolved in 50 ml of distilled water was added to the reaction mixture and the resultant mixture was heat-refluxed with stirring for 45 min. After cooling the mixture to room temperature, insoluble matters were filtered out, and the filtrate formed was distilled under reduced pressure. The residue was dissolved in ethyl acetate and the solution was washed with an aqueous 0.1N sodium hydroxide solution, dried by Glauber's salt, and purified by column chromatography using silica gel as fixed phase and a mixture of ethyl acetate and benzene (1/1) as moving phase. Thus, 30.5 g (yield 66%) of light yellow oily Intermediate 7 were obtained.

Synthesis of Intermediate 8

In a 300 mℓ three neck flask equipped with a reflux condenser and a stirrer were placed 30.5 g (0.072 mol) of Intermediate 7, 50 ml of acetonitrile, 14.1 g of p-toluenesulfonyl chloride, and 5.9 ml of pyridine and the mixture was stirred for 1h under room temperature. The disappearance of the Intermediate 7 by the reaction was confirmed by thin layer chromatography. Thereafter, 110 ml of methanol and 5.9 ml of pyridine were added thereto and the resultant mixture was heat-refluxed with stirring for 2 h. After adding thereto ethyl acetate, the mixture was washed with a saturated aqueous sodium chloride solution and supplied to column chromatography using a silica gel as fixed phase and a mixture of ethyl acetate and benzene (1/1) as moving phase to concentrate the product, whereafter it was crystallized by the addition of n-hexane. Crystals thus formed were filtered and dried to provide 6.6 g (yield 23%) of white crystals of Intermediate 8 having melting point of from 224 °C to 226 °C. The mass spectrum thereof showed a parent peak: m/e = 403.

Synthesis of Intermediate 9

In a 300 mℓ three neck flask equipped with a reflux condenser and a stirrer were placed 6.6 g (16.4 mmol) of Intermediate 8 and 100 ml of ethanol to form a suspension. While heat-refluxing the suspension with stirring, 0.10 g of $NaWO_4.H_2O$ and 13.5 ml of an aqueous 35% hydrogen peroxide solution were added thereto and the mixture was further heat-refluxed with stirring for 2 h. Then, the completion of the reaction was confirmed by thin layer chromatography. The reaction mixture was ice-cooled and crystals thus formed were collected by filtration, washed with ethanol, and then dried to provide 6.0 g (yield 84%) of the white crystals of Intermediate 9 having a melting point of 250 to 252 °C. Mass spectral parent peak: m/e = 435.

Synthesis of Intermediate 10

In a 300 mℓ three neck flask equipped with a reflux condenser and a stirrer were placed 6.0 g (14 mmol) of Intermediate 9 and 100 ml of isopropyl alcohol and while heat-refluxing the mixture (suspension) with stirring, 1.0 g of hydrazine hydrate was added dropwise to the suspension. Thereafter, the mixture was heat-refluxed with stirring for 3 h. The mixture was in the suspension state but the completion of the reaction was confirmed by thin layer chromatography. The reaction mixture was distilled under reduced pressure to remove the solvent, to form 6.6 g of solids. It was assumed that 4.2 g thereof were Intermediate 10 and 2.2 g thereof were phthalhydrazide.

Synthesis of Intermediate 12

In a 200 mℓ three neck flask equipped with a stirrer were placed 6.6 g of all of the aforesaid reaction mixture (i.e., Intermediate 10 and phthalhydrazide), 50 ml of N,N-dimethylacetamide, and 40 ml of

tetrahydrofuran and the reaction mixture was dissolved in the solvent mixture. Then, 9.7 g (14 mmol) of Intermediate 11 prepared by an ordinary manner were added to the solution with stirring at room temperature and further a solution of 1.9 ml of triethylamine dissolved in 10 ml of tetrahydrofuran was added dropwise to the mixture over a period of 10 min. The resultant mixture was stirred for 30 min at room temperature. In this case, the disappearance of raw material of Intermediate 11 was confirmed by thin layer chromatography, but since raw material of Intermediate 10 remained, 1.9 g of Intermediate 11 and 0.4 ml of triethylamine were additionally added to the mixture. The resultant mixture was stirred for 2 h at room temperature and the disappearance of Intermediate 10 was confirmed by thin layer chromatography. Then, after adding ethyl acetate, the reaction mixture was washed with a saturated aqueous sodium chloride solution of diluted hydrochloric acid, insoluble matter (presumed to be phthalhydrazide) was filtered off and the filtrate thus formed was purified by column chromatography using a silica gel as fixed phase and a mixture of ethyl acetate and chloroform (1/5) as moving phase to provide 7.7 g (yield 57% based on Intermediate 9) of the white crystals of Intermediate 12. The melting point thereof was from 87 °C to 94 °C.

Elemental Analysis:			
	H	C	N
Found:	7.48%	60.58%	8.60%
Calculated:	7.49%	60.72%	8.67%

Mass spectrum parent peak: m/e = 969 = $(M + H)^+$.

Synthesis of Compound 1

In a 300 mℓ three neck flask equipped with a stirrer were placed 2.5 g (2.6 mmol) of Intermediate 12, 65 ml of chloroform, and 50 ml of distilled water and while stirring the mixture at room temperature, 3.75 g of sodium carbonate, 0.81 g of mono-sulfate of 4-amino-3-methyl-N-ethyl-N-(β-methylsulfonamidoethyl)-aniline (D20 described above as specific example of aromatic primary amine), and 1.65 g of ammonium persulfate were successively added to the mixture. The reaction mixture was colored in blue to blue-green. After stirring the mixture for 1h at room temperature, the upper layer (aqueous layer) thus formed was removed by decantation, and the lower layer was washed well with an aqueous dilute acetic acid solution, purified by column chromatography using a silica gel as fixed phase and a mixture of methanol and chloroform (1/100) as moving phase, and further purified again by column chromatography using a silica gel as fixed phase and a mixture of ethyl acetate and chloroform (2/5) as moving phase. After subjecting the product to distillation under reduced pressure to dryness, the product was further dried for 10 h using a vacuum pump to provide 42.4 g (yield 75%) of amorphous Compound 1.

Elemental Analysis:			
	H	C	N
Found:	7.35%	59.24%	10.19%
Calculated:	7.25%	59.04%	10.13%

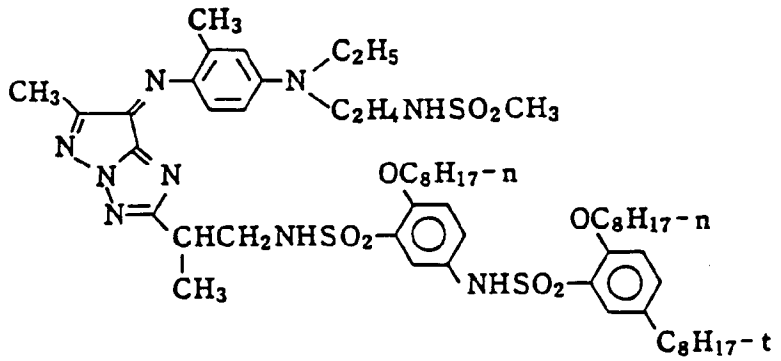
Mass spectrum parent peak: m/e = 1235 = $(M + H)^+$.

Example 2 Comparison of Visible Absorption Spectra

In a 100 mℓ measuring flask were placed 2.00 mg of Compound 1 (molecular weight = 1236.71) and after adding thereto ethyl acetate (high quality reagent) to dissolve the compound at room temperature, ethyl acetate was added up to the marked line. After light shaking the solution to form a homogeneous solution, the solution was placed in a quartz cell of 1 cm in thickness and the visible absorption spectrum thereof was measured by means of a ultraviolet visible ray spectrophotometer made by Shimazu Corporation.

Similarly, 3.00 mg of Comparison Compound A (molecular weight = 1110.56) shown below described in Japanese Patent Application (OPI) No. 186567/85 was dissolved in ethyl acetate and the visible absorption spectra was measured by the same manner as above.

Comparison compound A:



The visible absorption spectra of both are shown in Fig. 1 so standardized that the maximum absorption intensity becomes 1. The solid line shows the visible absorption spectra of Compound 1 of this invention and the dotted line shows that of comparison Compound A.

As is clear from Fig. 1, Compound 1 of this invention gives a sharper absorption spectrum than Comparison Example A, and the hue of the former is far superior to that of the latter. It can also be seen that the maximum absorption of the compound of this invention is shifted to the deep color side. That is, while Comparison Compound A shows a magenta color, Compound 1 of this invention shows a cyan color.

Furthermore, from the maximum absorption intensity and the mol density of each compound, the molecular extinction coefficient ϵ is calculated. These values are shown in Table 2.

Table 2

	Molecular Weight	Amount (mg)	Concentration (mol/ℓ)	Maximum Absorp. Intensity	Molecular Extinction Coefficient ϵ ($\ell.mol^{-1}.cm.^{-1}$)
Compound 1 of this Invention	1236.71	2.00	1.62×10^{-5}	1.438	8.89×10^{4}
Comparison Compound A	1110.56	2.00	1.81×10^{-5}	1.005	5.58×10^{4}

From the results shown in the above table, it can be seen that Compound 1 of this invention has a significantly higher molecular extinction coefficient compared to Comparison Compound A, and hence Compound 1 of this invention provides the desired optical density in a smaller amount.

From the above-described results, it can be seen that when a particularly strong electron attractive group (the σ_p value of benzenesulfonyl group which is a substituent of Compound 1 of this invention is 0.70, while the σ_p value of the methyl group which is the substituent at the 6-position of Comparison Compound A is -0.17) is introduced to the substituent of the pyrazoloazole skeleton of the pyrazoloazoleazomethine dye, the maximum absorption wavelength is greatly shifted to a deep color side, the absorption thereof becomes sharper to make the hue clearer, and the molecular extinction coefficient becomes significantly larger.

Example 3 Synthesis of Compound 2

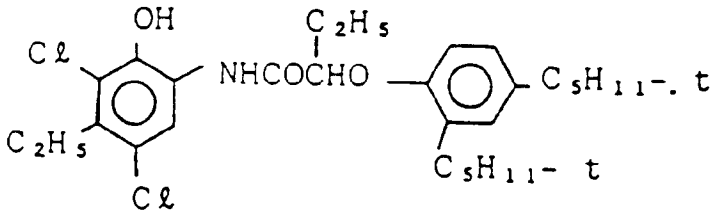
By following the same synthesis route as in Example 1 using Intermediate 3 and trifluoromethane thiol, amorphous Compound 2 was obtained.

Elemental Analysis:						
	H	C	N	F	Cl	S
Found:	6.53%	53.47%	8.33%	5.50%	3.45%	9.61%
Calculated:	6.68%	53.08%	8.44%	5.72%	3.56%	9.66%

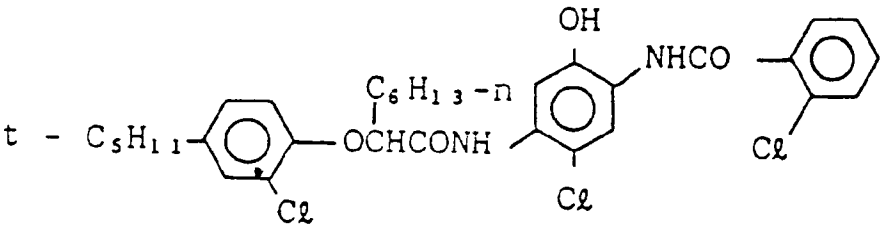
Mass spectrum parent peak: m/e = 995 = $(M + H)^+$.

Example 4 Comparison with Cyan Dye Formed from Known Cyan Coupler

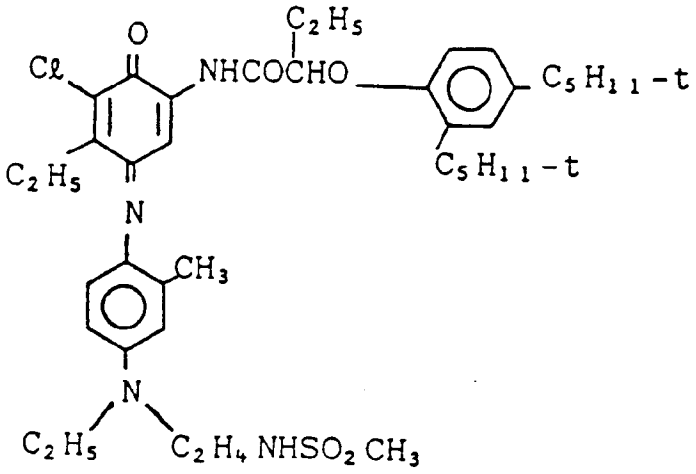
By following the same procedure as in Example 1, cyan dyes (indoaniline dyes) D-3 and D-4 were prepared from cyan couplers C-3 disclosed in U.S. Patent 3,772,002 and C-4 disclosed in U.S. Patent 4,560,635, respectively, shown below.



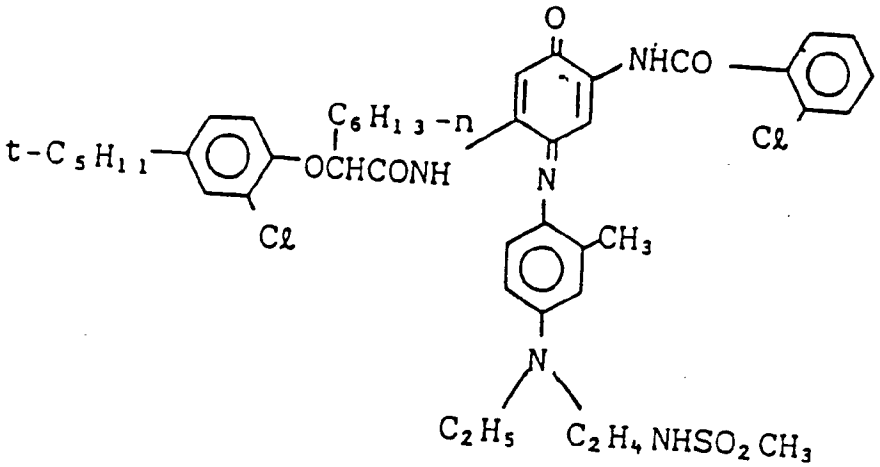
C - 3



C - 4



D - 3



D - 4

In the same manner as shown in Example 2, the visible absorption spectrum and the molecular extinction coefficient of each dye thus formed were measured, and they were compared with those of Compound 1 of this invention. The visible absorption spectra of these dyes are shown in Fig. 2, wherein the visible absorption spectrum of the Compound 1 of this invention is shown by a solid line. As shown in Figure 2, the absorption spectrum of Compound 1 of this invention is much sharper than that of comparison dyes D-3 and D-4, and Compound 1 scarcely has an absorption in a blue region of from 400 n.m. to 500 n.m., hence, the hue of the dye of this invention is a very clear cyan color in spite of the fact that the maximum absorption wavelength of Compound 1 of this invention is shorter than those of comparison dyes D-3 and D-4.

The data relating to the molecular extinction coefficients of these dyes are shown in Table 3.

Table 3

	Molecular Weight	Amount (mg)	Concentration (mol/ℓ)	Maximum Absorp. Intensity	Molecular Extinction Coefficient ϵ ($\ell.mol^{-1}.cm.^{-1}$)
Compound 1 of this Invention	1236.71	2.00	1.62×10^{-5}	1.438	8.89×10^{4}
Comparison Dye D-3	741.42	2.00	2.70×10^{-5}	0.661	2.45×10^{4}
Comparison Dye D-4	852.91	2.00	2.34×10^{-5}	0.551	2.35×10^{4}

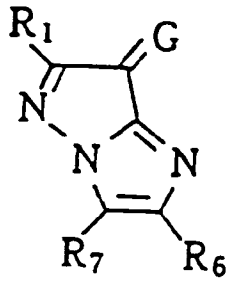
From the results shown in Table 3, it can be seen that Compound 1 of this invention gives a significantly higher molecular extinction coefficient compared to comparison dyes D-3 and D-4.

From the results described above, it can be seen that the image-forming cyan dyes formed from the cyan coupler of this invention gives a very sharp absorption spectrum compared with cyan dyes formed from conventional phenolic cyan couplers and hence gives a clear cyan hue as well as showing a very high molecular extinction coefficient. Thereby, a desired optical density can be obtained with a very small amount of the dye.

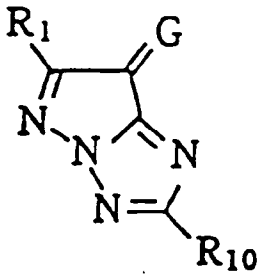
Claims

1. A pyrazoloazoleazomethine dye represented by formulae (II), (IV), (V), (VI) or (VII),

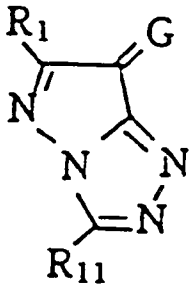
(II)



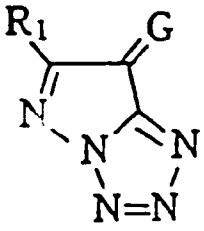
(IV)



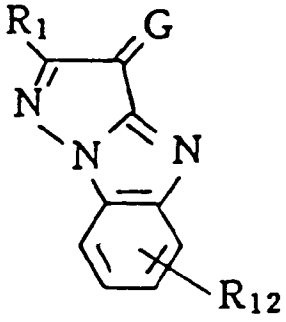
(V)



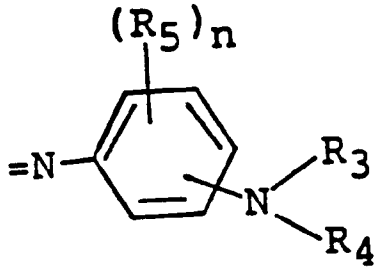
(VI)



(VII)



wherein G represents a structural moiety represented by formula (VIII)



(VIII)

wherein R_1 represents a substituent having a Hammett's substituent constant value of at least 0.6; R_3 and R_4 each represents a hydrogen atom or a substituted or unsubstituted alkyl group and R_5 represents a hydrogen atom, a chlorine or a substituted or unsubstituted alkyl group; n represents 1 or 2;

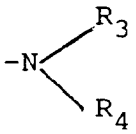
R_6, R_7, R_{10}, R_{11} and R_{12} each represents a hydrogen atom, a halogen atom, an alkyl group, an aralkyl group, an alkenyl group, an alkynyl group, a cycloalkyl group, a cycloalkenyl group, an aryl group, a heterocyclic group, a cyano group, an alkoxy group, an aryloxy group, an acylamino group, an anilino group, a ureido group, a sulfamoylamino group, an alkylthio group, an arylthio group, an alkoxycarbonylamino group, a sulfonamido group, a carbamoyl group, a sulfamoyl group, a sulfonyl group, an alkoxycarbonyl group, a heterocyclic oxy group, an acyloxy group, a carbymoyloxy group, a silyloxy group, an aryloxycarbonylamino group, an imido group, a heterocyclic thio group, a sulfinyl group, a phosphonyl group, an aryloxycarbonyl group or an acyl group;

or said dye being in the form of a dimer or higher polymer by combining with each other or to a polymer through a divalent group at R_1, R_5, R_7, R_{10}, R_{11} or R_{12}.

2. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein R_1 is a cyano group, a nitro group a trialklammonium group, a triarylammonium group, a dialkylsulfonium group, a diarylsulfonium group, a perfluoroalkylsulfinyl group, an ω-hydroperfluoroalkylsulfinyl group, an alkanesulfonyl group, an arylsulfonyl group, a β-carboxyvinyl group, or a β,β-dicyanovinyl group.

3. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein R_1 is a cyano group, a β-carboxyvinyl group, a nitro group, a trimethylammonium group, a trifluoromethanesulfinyl group, a trifluoromethanesulfonyl group, a difluoromethanesulfonyl group, a methanesulfonyl group, a dimethylsulfonium group, a benzenesulfonyl group, or a β,β-dicyanovinyl group.

4. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein R_5 and



are at the o- and p-position of the benzene ring, respectively.

5. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein the dye is represented by formula (IV), (V), or (VI).

6. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein the dye is represented by formula (IV).

7. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein said substituents represented by R_6, R_7, R_{10}, R_{11} or R_{12} each represents a hydrogen atom, a halogen atom, an alkyl group, an aryl group, a heterocyclic group, a cyano group, an alkoxy group, an aryloxy group, an acylamino group, an anilino group, a ureido group, a sulfamoylamino group, an alkylthio group, an arylthio group, an alkoxycarbonylamino group, a sulfonamido group, a carbamoyl group, a sulfamoyl group, a sulfonyl group, an alkoxycarbonyl group, a heterocyclic oxy group, an acyloxy group, a carbomoyloxy group, a silyloxy

group, an aryloxycarbonylamino group, an imido group, a heterocyclic thio group, a sulfinyl group, a phosphonyl group, an aryloxycarbonyl group, or an acyl group.

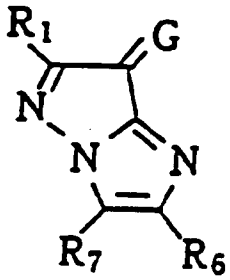
8. The pyrazoloazoleazomethine dye as claimed in claim 1 wherein the substituent for the alkyl group represented by R_3 or R_4 is a hydroxyl group, an alkoxy group, an alkoxyalkoxy group or an alkylsulfonamido group.

9. The pyrazoloazoleazomethine dye as claimed in claim 1, wherein the substituent for the alkyl group represented by R_5 is an alkoxy group or a halogen atom.

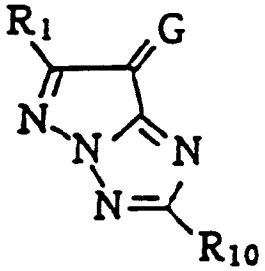
Patentansprüche

1. Pyrazoloazolazomethinfarbstoff, dargestellt durch die Formeln (II), (IV), (V), (VI) oder (VII)

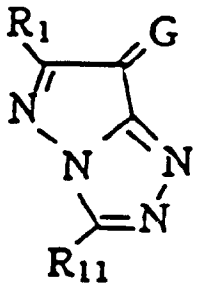
(II)



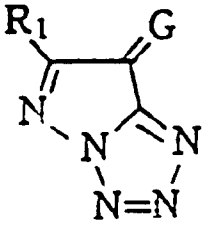
(IV)



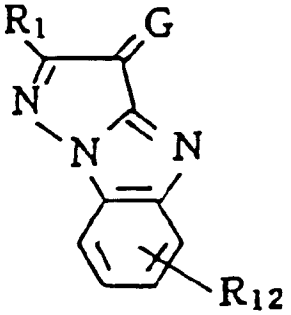
(V)



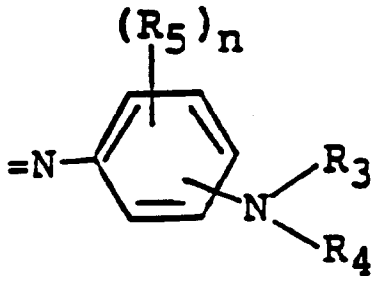
(VI)



(VII)



worin G eine Struktureinheit, dargestellt durch die Formel (VIII), darstellt



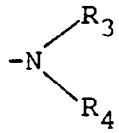
(VIII)

worin R_1 einen Substituenten mit einem Wert der Hammett-Substituentenkonstante von mindestens 0,6 darstellt;
R_3 und R_4 jeweils ein Wasserstoffatom oder eine substituierte oder unsubstituierte Alkylgruppe darstellen und R_5 ein Wasserstoffatom, Chlor oder eine substituierte oder unsubstituierte Alkylgruppe darstellt; n 1 oder 2 ist;
R_6, R_7, R_{10}, R_{11} und R_{12} jeweils ein Wasserstoffatom, ein Halogenatom, eine Alkylgruppe, eine Aralkylgruppe, eine Alkenylgruppe, eine Alkynylgruppe, eine Cycloalkylgruppe, eine Cycloalkenylgruppe, eine Arylgruppe, eine heterocyclische Gruppe, eine Cyanogruppe, eine Alkoxygruppe, eine Aryloxygruppe, eine Acylaminogruppe, eine Anilinogruppe, eine Ureidogruppe, eine Sulfamoylaminogruppe, eine Alkylthiogruppe, eine Arylthiogruppe, eine Alkoxycarbonylaminogruppe, eine Sulfonamidogruppe, eine Carbamoylgruppe, eine Sulfamoylgruppe, eine Sulfonylgruppe, eine Alkoxycarbonylgruppe, eine heterocyclische Oxygruppe, eine Acyloxygruppe, eine Carbymoyloxygruppe, eine Silyloxygruppe, eine Aryloxycarbonylaminogruppe, eine Imidogruppe, eine heterocyclische Thiogruppe, eine Sulfinylgruppe, eine Phosphonylgruppe, eine Aryloxycarbonylgruppe oder eine Acylgruppe darstellen;
oder der Farbstoff liegt in Form eines Dimeren oder höheren Polymers durch Kombination miteinander oder mit einem Polymer über eine divalente Gruppe an R_1, R_6, R_7, R_{10}, R_{11} oder R_{12} vor.

2. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin R_1 eine Cyanogruppe, eine Nitrogruppe, eine Trialkylammoniumgruppe, eine Triarylammoniumgruppe, eine Dialkylsulfoniumgruppe, eine Diarylsulfoniumgruppe, eine Perfluoroalkylsulfinylgruppe, eine ω-Hydroperfluoroalkylsulfinylgruppe, eine Alkansulfonylgruppe, eine Arylsulfonylgruppe, eine β-Carboxyvinylgruppe oder eine β,β-Dicyanovinylgruppe ist.

3. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin R_1 eine Cyanogruppe, eine β-Carboxyvinylgruppe, eine Nitrogruppe, eine Trimethylammoniumgruppe, eine Trifluoromethansulfinylgruppe, eine Trifluoromethansulfonylgruppe, eine Difluoromethansulfonylgruppe, eine Methansulfonylgruppe, eine Dimethylsulfoniumgruppe, eine Benzolsulfonylgruppe oder eine β,β-Dicyanovinylgruppe ist.

4. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin R_5 und



an der o- beziehungsweise an der p-Position des Benzolrings sind.

5. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin der Farbstoff durch die Formel (IV), (V) oder (VI) dargestellt ist.

6. Pyrazolazoloazomethinfarbstoff nach Anspruch 1, worin der Farbstoff durch die Formel (IV) dargestellt ist.

7. Pyrazolazoloazomethinfarbstoff nach Anspruch 1, worin jeder der Substituenten, dargestellt durch R_6, R_7, R_{10}, R_{11} oder R_{12}, ein Wasserstoffatom, ein Halogenatom, eine Alkylgruppe, eine Arylgruppe, eine heterocyclische Gruppe, eine Cyanogruppe, eine Alkoxygruppe, eine Aryloxygruppe, eine Acylaminogruppe, eine Anilinogruppe, eine Ureidogruppe, eine Sulfamoylaminogruppe, eine Alkylthiogruppe, eine Arylthiogruppe, eine Alkoxycarbonylaminogruppe, eine Sulfonamidogruppe, eine Carbamoylgruppe, eine Sulfamoylgruppe, eine Sulfonylgruppe, eine Alkoxycarbonylgruppe, eine heterocyclische Oxygruppe, eine Acyloxygruppe, eine Carbomoylgruppe, eine Silyloxygruppe, eine Aryloxycarbonylaminogruppe, eine Imidogruppe, eine heterocyclische Thiogruppe, eine Sulfinylgruppe, eine Phosphonylgruppe, eine Aryloxycarbonylgruppe oder eine Acylgruppe darstellt.

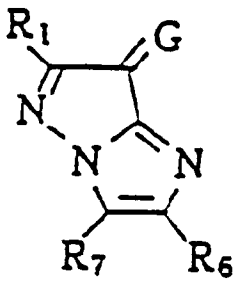
8. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin der Substituent für die Alkylgruppe, dargestellt durch R_3 oder R_4, eine Hydroxylgruppe, eine Alkoxygruppe, eine Alkoxyalkoxygruppe oder eine Alkylsulfonamidogruppe ist.

9. Pyrazoloazolazomethinfarbstoff nach Anspruch 1, worin der Substituent für die Alkylgruppe, dargestellt durch R_5, eine Alkoxygruppe oder ein Halogenatom ist.

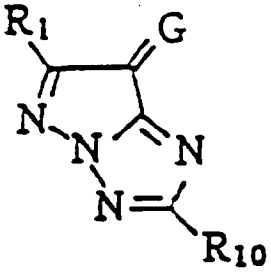
Revendications

1. Colorant pyrazoloazoleazométhinique représenté par les formules (II), (IV), (V), (VI) ou (VII),

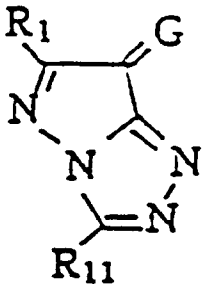
(II)



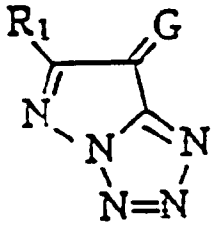
(IV)



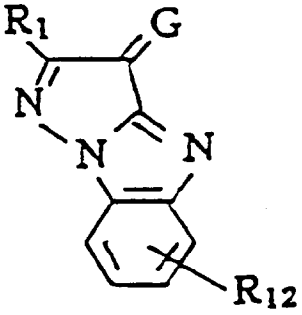
(V)



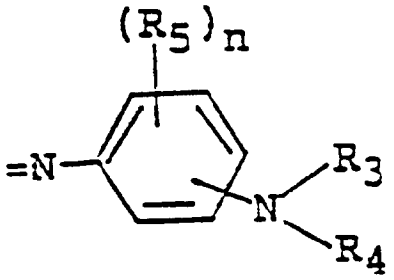
(VI)



(VII)



dans lesquelles G représente une portion structurale représentée par la formule (VIII)



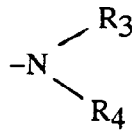
(VIII)

formules dans lesquelles R_1 représente un substituant ayant une valeur de la constante de substituant de Hammett d'au moins 0,6,
R_3 et R_4 représentent chacun un atome d'hydrogène ou un groupe alkyle substitué ou non substitué et R_5 représente un atome d'hydrogène, un atome de chlore ou un groupe alkyle substitué ou non substitué, n représente 1 ou 2,
R_6, R_7, R_{10}, R_{11} et R_{12} représentent chacun un atome d'hydrogène, un atome d'halogène, un groupe alkyle, un groupe aralkyle, un groupe alcényle, un groupe alcynyle, un groupe cycloalkyle, un groupe cycloalcényle, un groupe aryle, un groupe hétérocyclique, un groupe cyano, un groupe alcoxy, un groupe aryloxy, un groupe acylamino, un groupe anilino, un groupe uréido, un groupe sulfamoylamino, un groupe alkylthio, un groupe arylthio, un groupe alcoxycarbonylamino, un groupe sulfonamido, un groupe carbamoyle, un groupe sulfamoyle, un groupe sulfonyle, un groupe alcoxycarbonyle, un groupe oxy hétérocyclique, un groupe acyloxy, un groupe carbamoyloxy, un groupe sylyloxy, un groupe aryloxycarbonylamino, un groupe imido, un groupe thio hétérocyclique, un groupe sulfinyle, un groupe phosphonyle, un groupe aryloxycarbonyle ou un groupe acyle, ou ledit colorant est sous forme d'un dimère ou d'un polymère supérieur par combinaison entre eux ou en un polymère par un groupe divalent au niveau de R_1, R_6, R_7, R_{10}, R_{11} ou R_{12}.

2. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel R_1 est un groupe cyano, un groupe nitro, un groupe trialkylammonium, un groupe triarylammonium, un groupe dialkylsulfonium, un groupe diarylsulfonium, un groupe perfluoroalkylsulfinyle, un groupe ω-hydroperfluoroalkylsulfinyle, un groupe alcanesulfonyle, un groupe arylsulfonyle, un groupe β-carboxyvinyle ou un groupe β,β-dicyanovinyle.

3. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel R_1 est un groupe cyano, un groupe β-carboxyvinyle, un groupe nitro, un groupe triméthylammonium, un groupe trifluorométhanesulfinyle, un groupe trifluorométhanesulfonyle, un groupe difluorométhanesulfonyle, un groupe méthanesulfonyle, un groupe diméthylsulfonium, un groupe benzènesulfonyle ou un groupe β,β-dicyanovinyle.

4. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel R_5 et



sont en position o et p du cycle benzénique, respectivement.

5. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel le colorant est représenté par la formule (IV), (V) ou (VI).

6. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel le colorant est représenté par la formule (IV).

7. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel lesdits substituants représentés par R_6, R_7, R_{10}, R_{11} ou R_{12} représentent chacun un atome d'hydrogène, un atome d'halogène,

un groupe alkyle, un groupe aryle, un groupe hétérocyclique, un groupe cyano, un groupe alcoxy, un groupe aryloxy, un groupe acylamino, un groupe anilino, un groupe uréido, un groupe sulfamoylamino, un groupe alkythio, un groupe arylthio, un groupe alcoxycarbonylamino, un groupe sulfonamido, un groupe carbamoyle, un groupe sulfamoyle, un groupe sulfonyle, un groupe alcoxycarbonyle, un groupe oxy hétérocyclique, un groupe acyloxy, un groupe carbamoyloxy, un groupe silyloxy, un groupe aryloxycarbonylamino, un groupe imido, un groupe thio hétérocyclique, un groupe sulfinyle, un groupe phosphonyle, un groupe aryloxycarbonyle ou un groupe acyle.

8. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel le substituant du groupe alkyle représenté par R_3 ou R_4 est un groupe hydroxyle, un groupe alcoxy, un groupe alcoxyalcoxy ou un groupe alkylsulfonamido.

9. Colorant pyrazoloazoleazométhinique selon la revendication 1, dans lequel le substituant du groupe alkyle représenté par R_5 est un groupe alcoxy ou un atome d'halogène.

Fig. 1

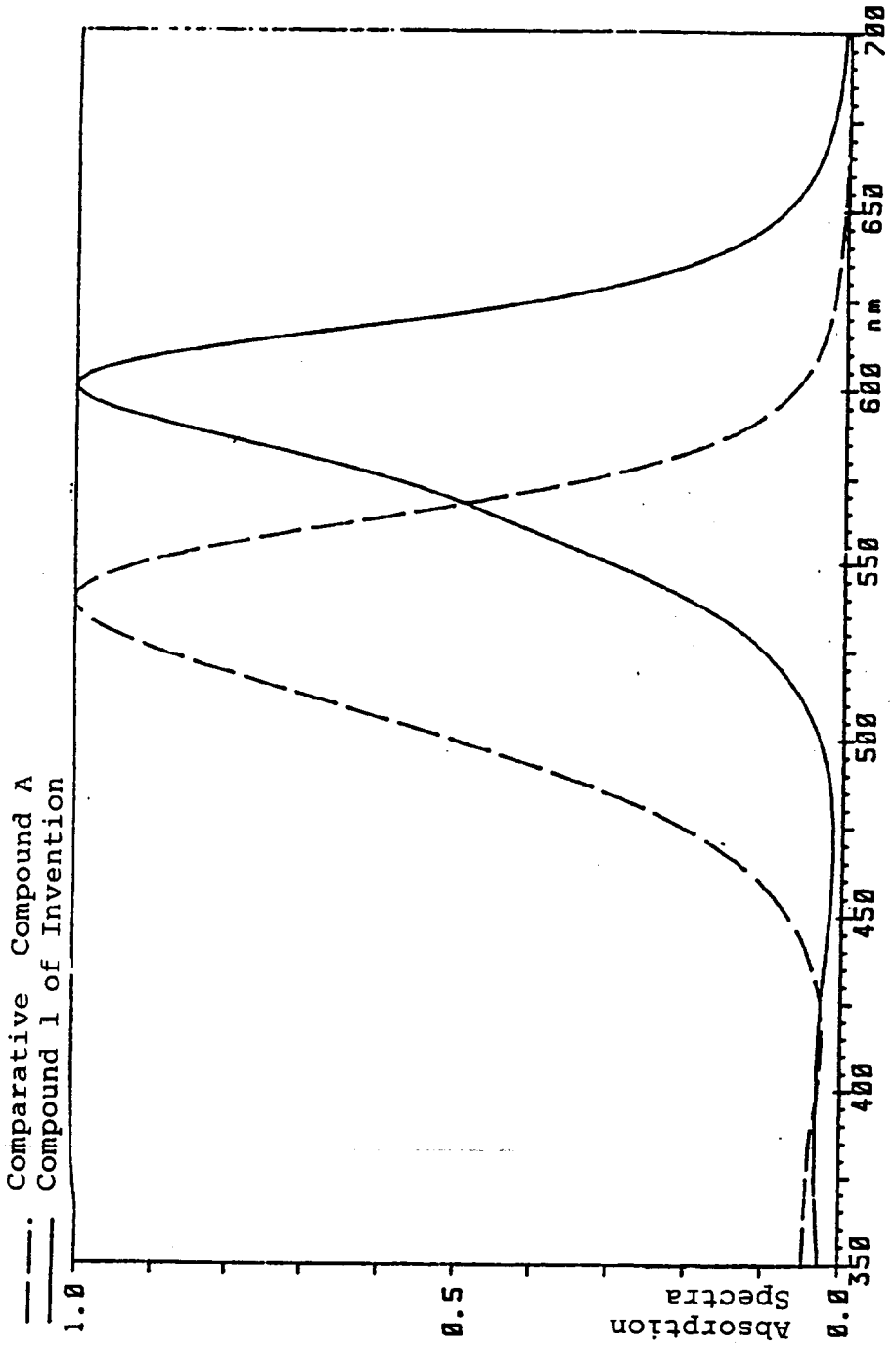
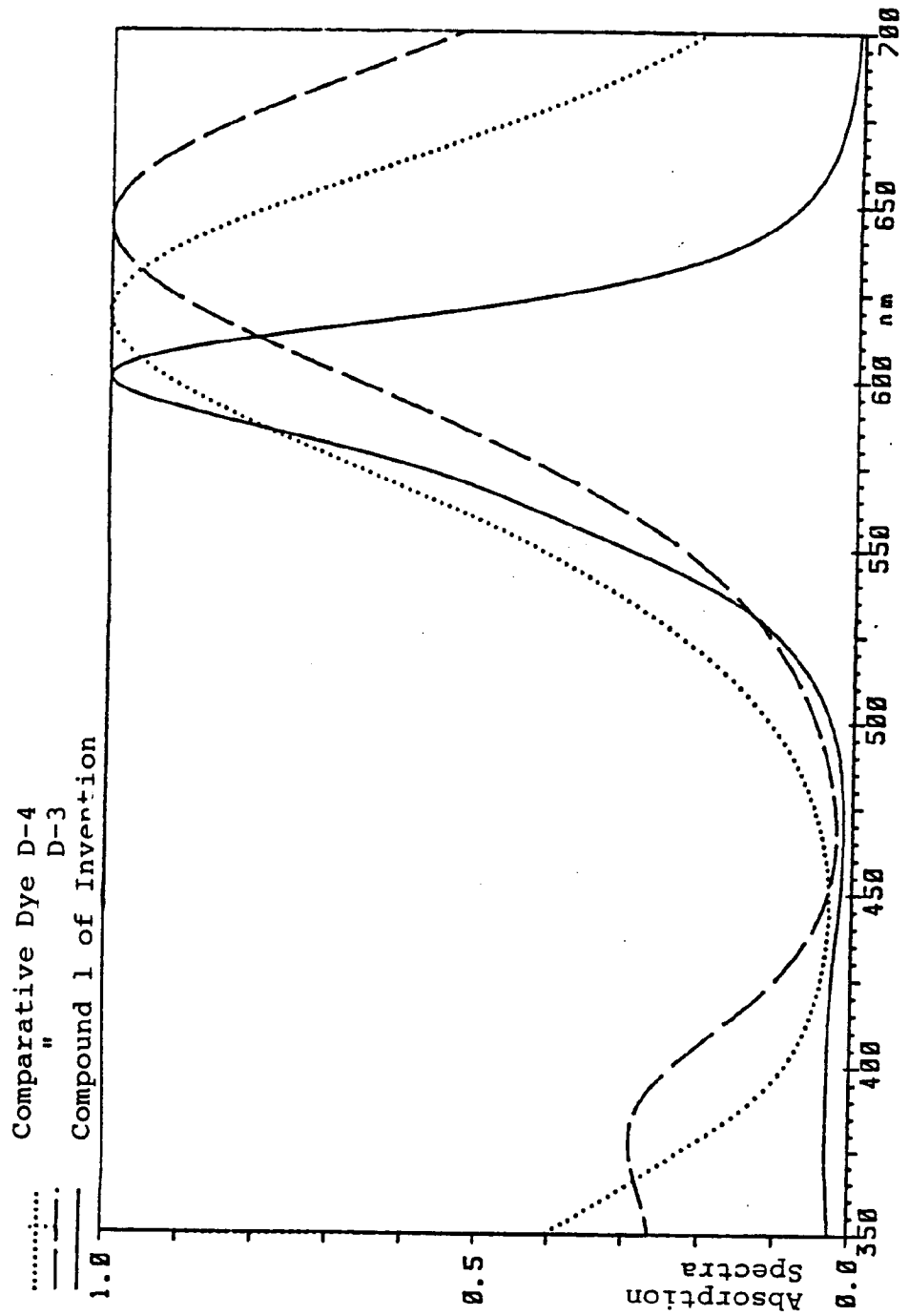

Comparative Compound A
Compound 1 of Invention
Absorption Spectra
1.0
0.5
0.0
350
400
450
500
550
600
650
700
nm

Fig. 2


Comparative Dye D-4
" D-3
Compound 1 of Invention
1.0
0.5
0.0
Absorption Spectra
350
400
450
500
550
600
650
700
nm